

Influence of nitrate radical on the oxidation of dimethyl sulfide in a polluted marine environment

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DMS is the most important (90%) contributor to biogenic atmospheric sulfur. Its contribution to total atmospheric sulfur varies, depending on anthropogenic emissions. It is a major contributor to total sulfur in the southern hemisphere. The chemical DMS oxidation mechanism determines the aerosol yield. If the H2SO4 yield from DMS oxidation by NO3 is different, then NO3 could have an influence on aerosol production.

What happens to DMS in the marine boundary layer (MBL) determines the global role of DMS and its role in the free troposphere.

Questions: (1) Can we see a clear role of NO₃ in DMS oxidation?

(2) What is the contribution of NO₃ to the removal of DMS from MBL? (3) What is the contribution of DMS to conversion of NO₃ to HNO₃, and hence removal of NO. in MBL?

NO3 Detection: Cavity Ringdown Spectroscopy



Optical detection @662 nm Measure light decay rate from optical cavity Empty cavity ringdown times ≈ 200 µs = path length of 60 km Dual cell instrument for simultaneous detection of NO2 and N2OE Sample rate = 20 Hz, averaged for 1 min data More details: see poster A51D-0716 Detection limit: 1 pptv

0.1

DMS detection: · Gas Chromatography/Mass Spectrometry • 5 Minutes sampling time; repeat every 30 minutes, detection limit = 0.5 pptv



Diurnal averages of DMS, NO₃, and OH, comparison with chemical box model 100 [OH], 80×[CI] (10⁶ 80 (pptv) 60 ő molecules DMS, 4∩ 20 B DMS, meas OH, o DMS, DMS, calc. 5, model, r 0................ ******** N 5 10 15 20 0 Local Time (hours) Good agreement between modeled and measured DMS during most periods of the day

- NO₃ oxidizes $\approx 80\%$ of the available DMS in comparison to OH in the coastal region
- Inclusion of Cl as a proxy for halogens improves daytime agreement

Conclusions

- DMS and NO₃ anticorrelated, clear indication of NO₃ reaction with DMS • DMS oxidation by NO₃ remains important well downwind of NO₈ sources
- DMS oxidation by NO₃ 4 times more important than by OH during NEAOS
- DMS oxidation by NO₃ can be equal to oxidation by OH up to 3000 km downwind
- Aerosol production and growth altered by NO₃-DMS oxidation
- The flux of DMS out of the MBL will be suppressed in NO₃-rich regimes
- Reaction with DMS is one of the most important pathways for NO, removal
- DMS + NO₂ is an important contribution to nocturnal nitric acid production

These results have been recently published:

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100

10

DMS (pptv)