



# **Chemical Characterization of Water Soluble Organic Aerosol in Contrasting Rural and Urban Environments in the Southeastern United States**

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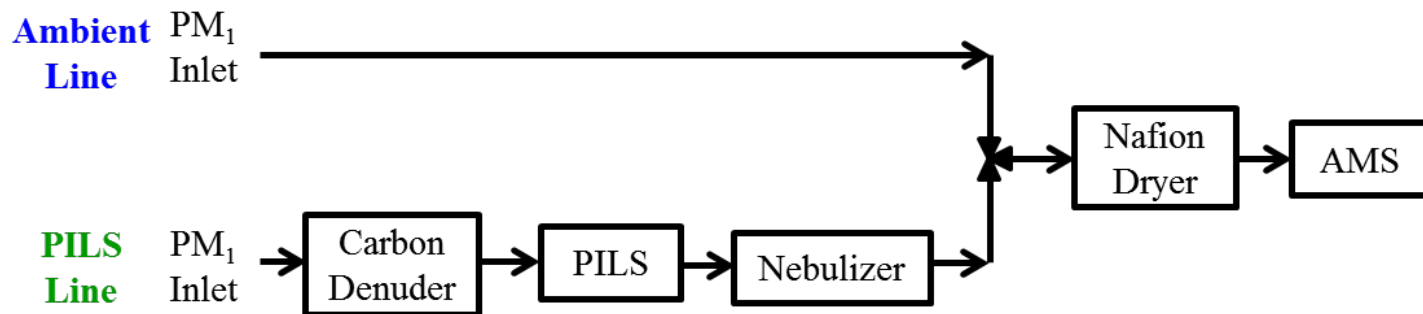


**AMS Users Meeting, 2017**

# Introduction

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- Water-soluble organic matter (WSOM) is an important component of atmospheric PM
  - alter the hygroscopicity and surface tension of particles
  - chemical composition and sources of WSOM are poorly understood
- Water-solubility of organic aerosols from different sources is different
  - possible to understand the sources of atmospheric OA from the perspective of water-solubility



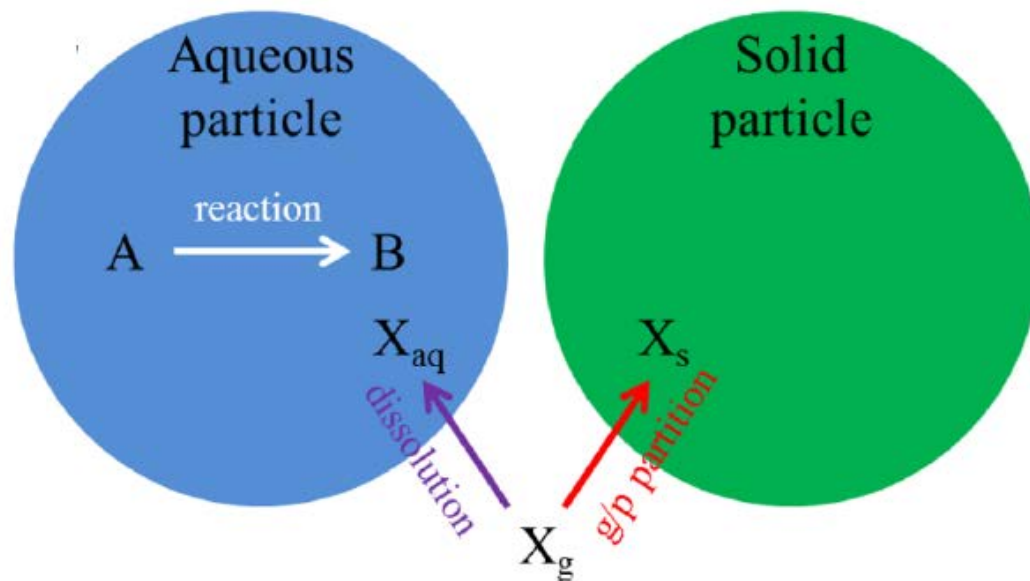
- Online collection and quantification of water-soluble OA and total ambient OA with one instrument
- The AMS alternated sampling between ambient line and PILS line every 30 min.

SOAS: Rural site in Centreville, AL (June 1- July 15, 2013)

Post-SOAS: Urban site in Atlanta, GA (August, 2013)

# Note: Species Collected by PILS

Xu et al., ES&T, 2017

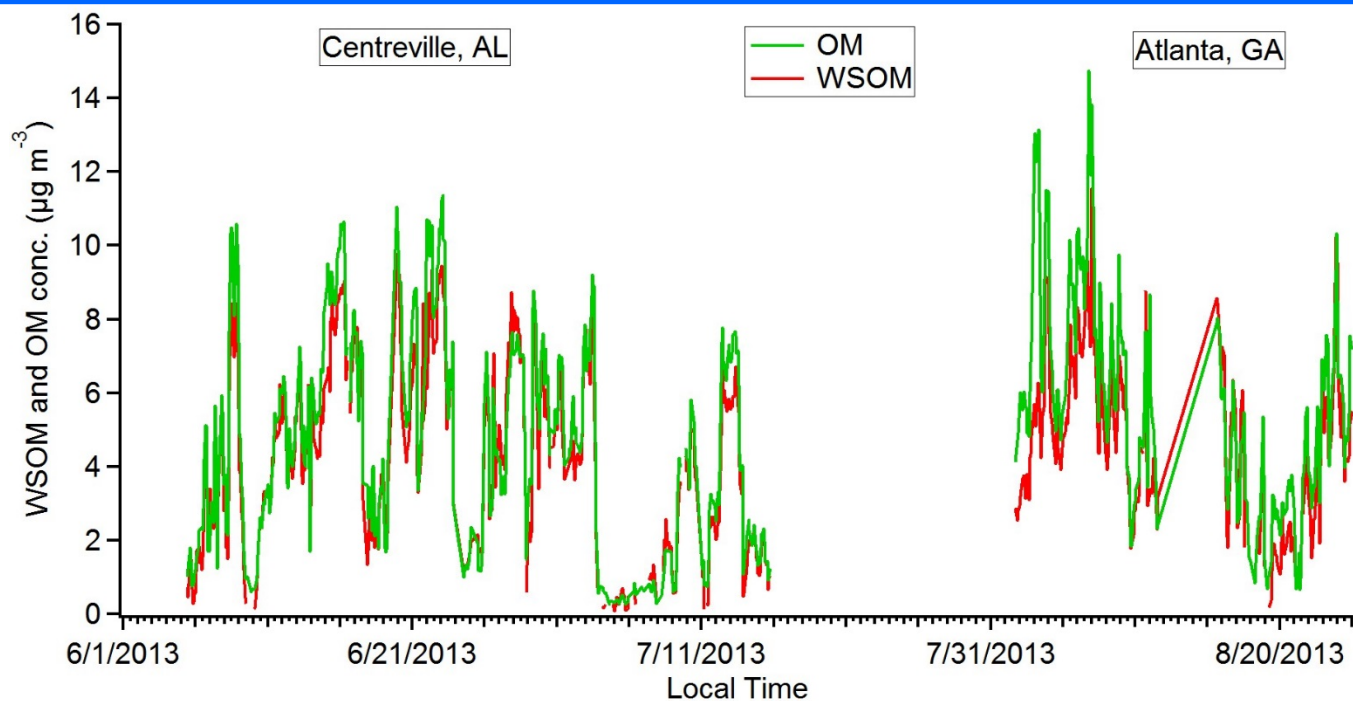


**Species measured by PILS: B,  $X_{aq}$ , and  $X_s$**

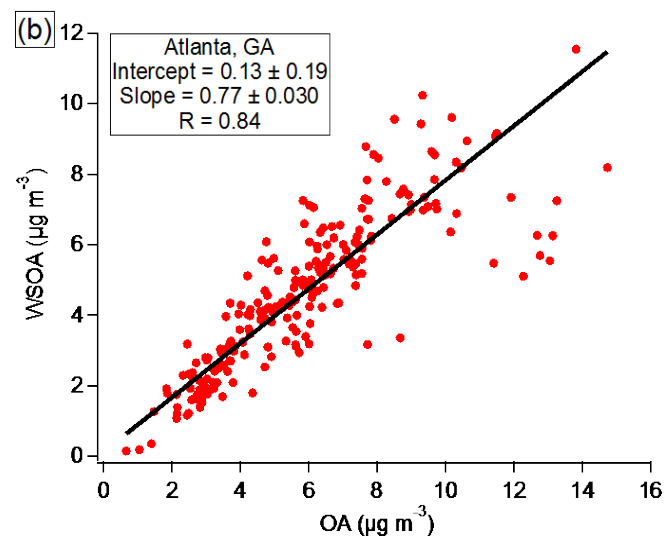
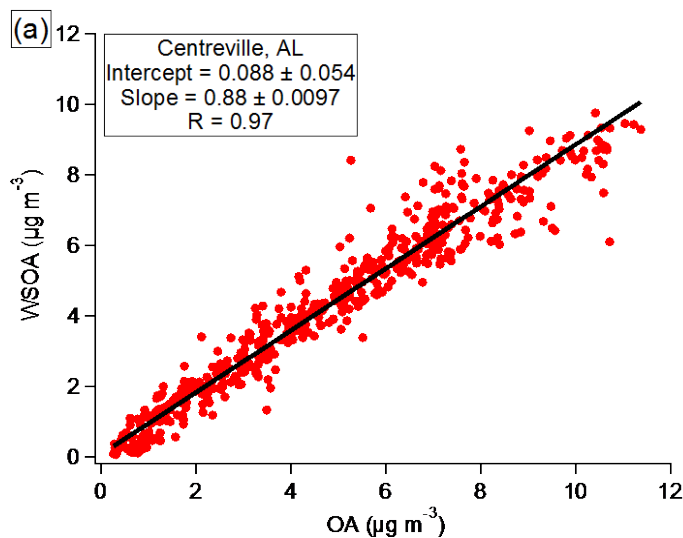
**NOT THE SAME AS**

**Species formed through aqueous-phase reactions (i.e., B) or  
Species already in the ambient aqueous aerosols (i.e., B and  $X_{aq}$ )**

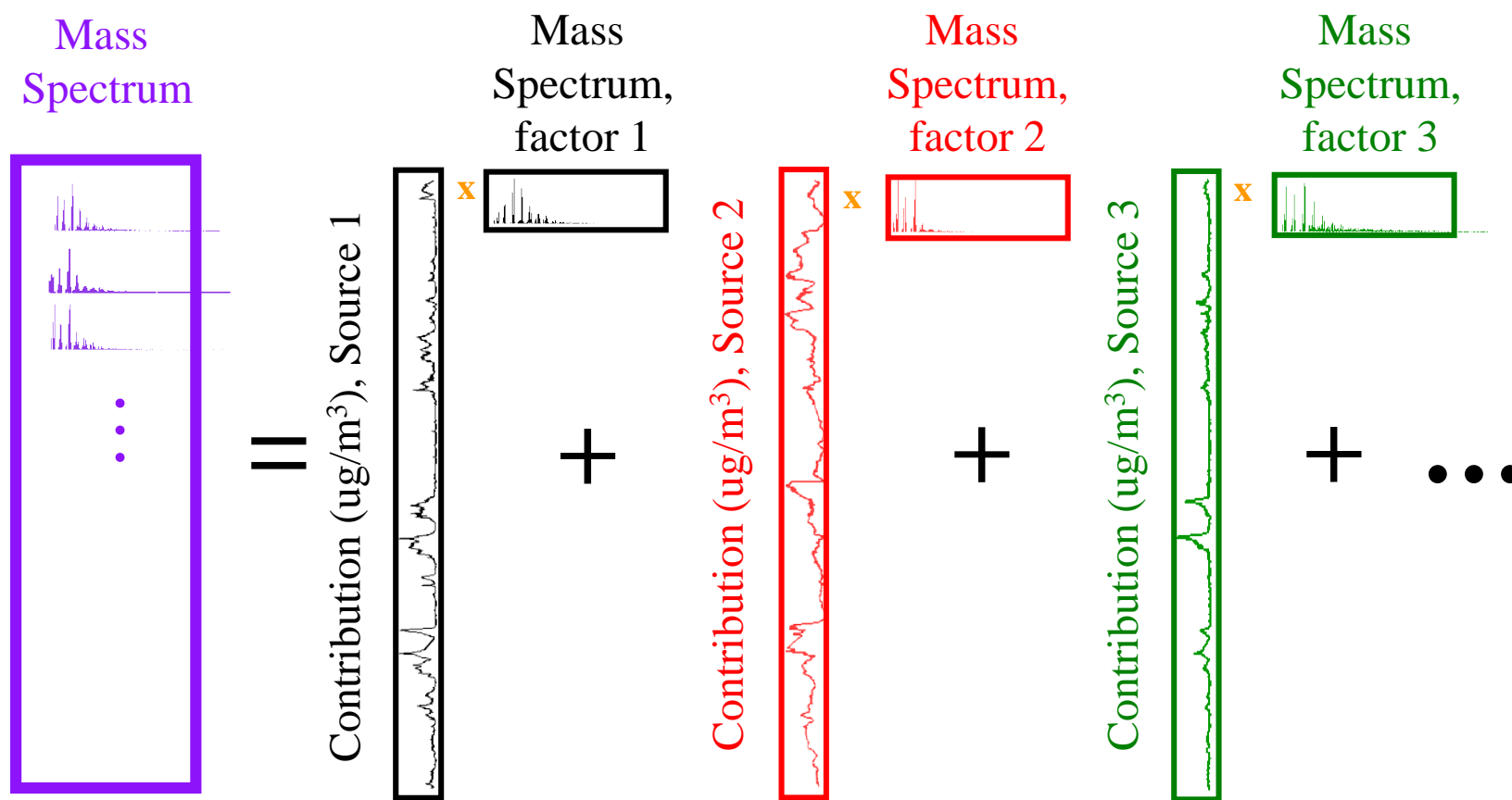
# Water-solubility of Organic Aerosols



Xu et al.,  
ES&T, 2017



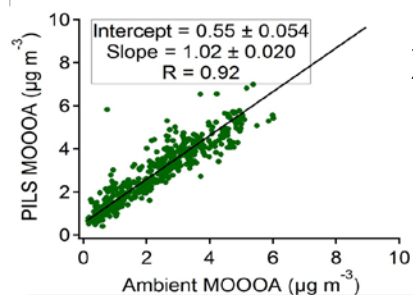
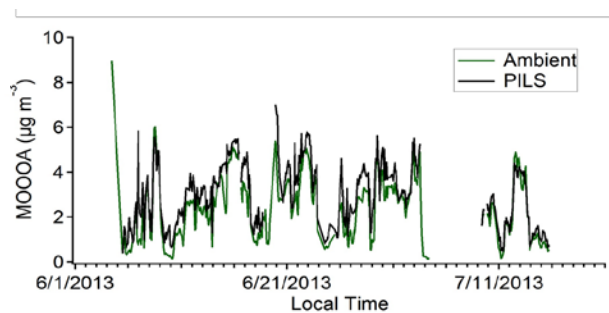
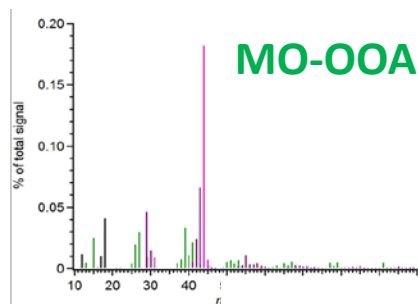
# Positive Matrix Factorization (PMF) analysis



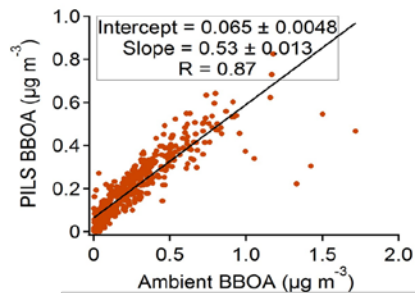
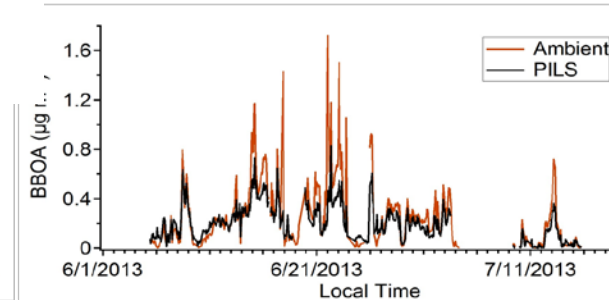
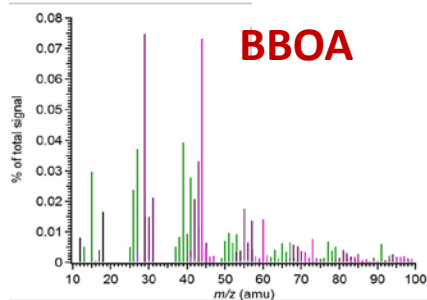
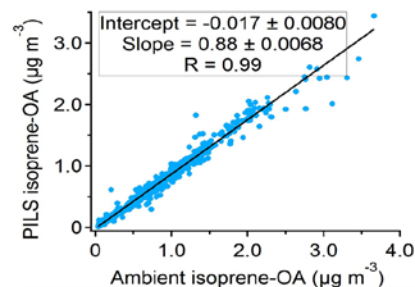
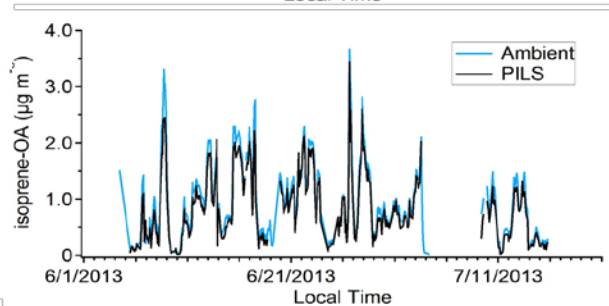
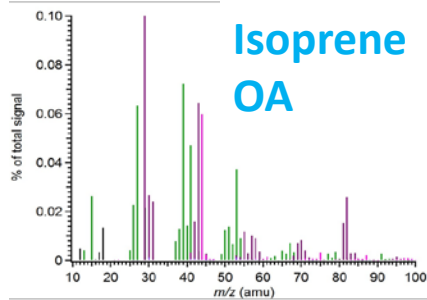
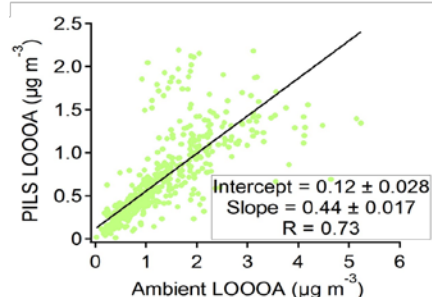
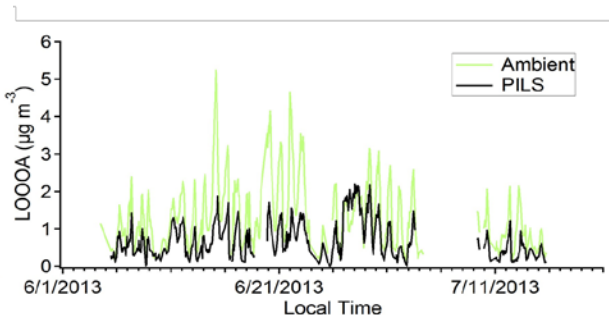
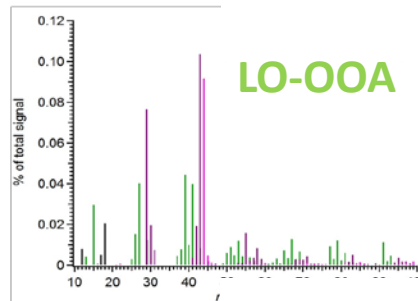
PMF is a bilinear mixing model in which a dataset matrix is assumed to be comprised of the linear combination of factors with constant profiles that have varying contributions across the dataset.



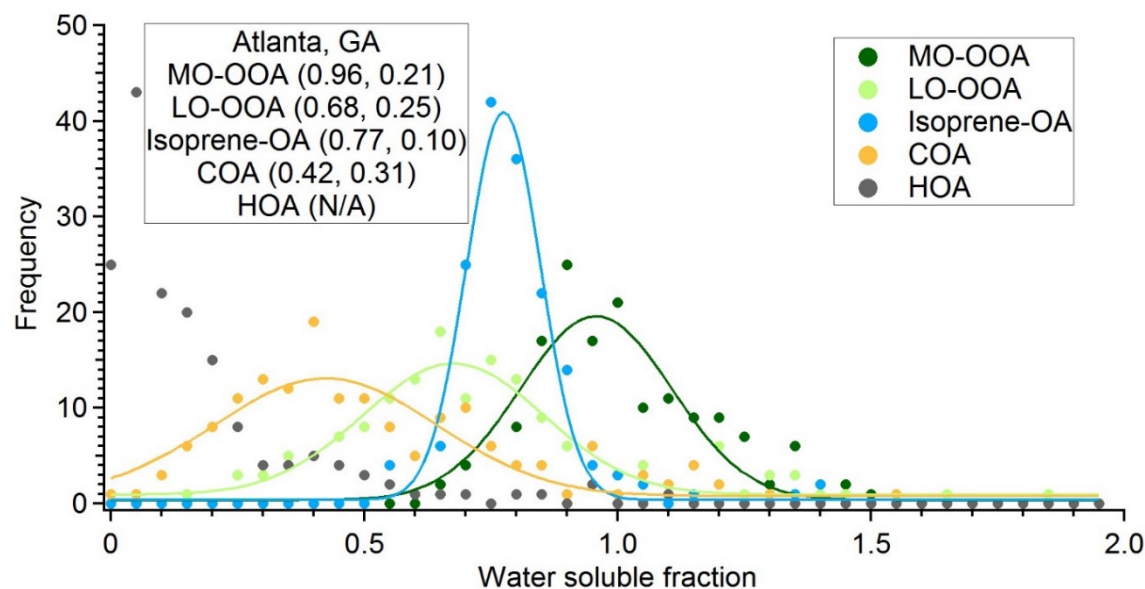
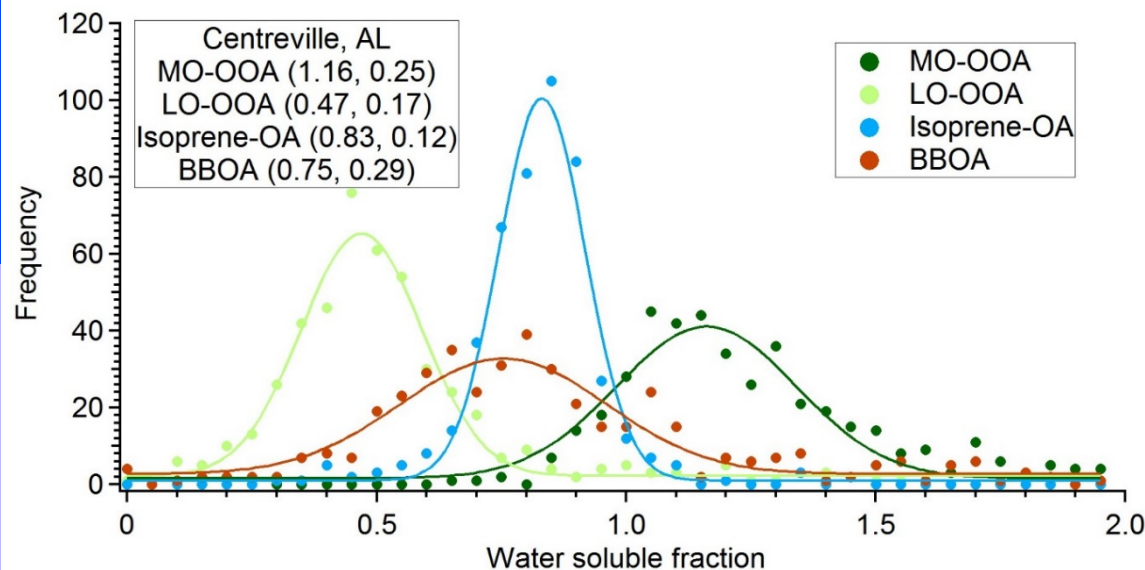
# Water-Solubility of OA Factors



Xu et al.,  
ES&T, 2017



# Water-Solubility of OA Factors

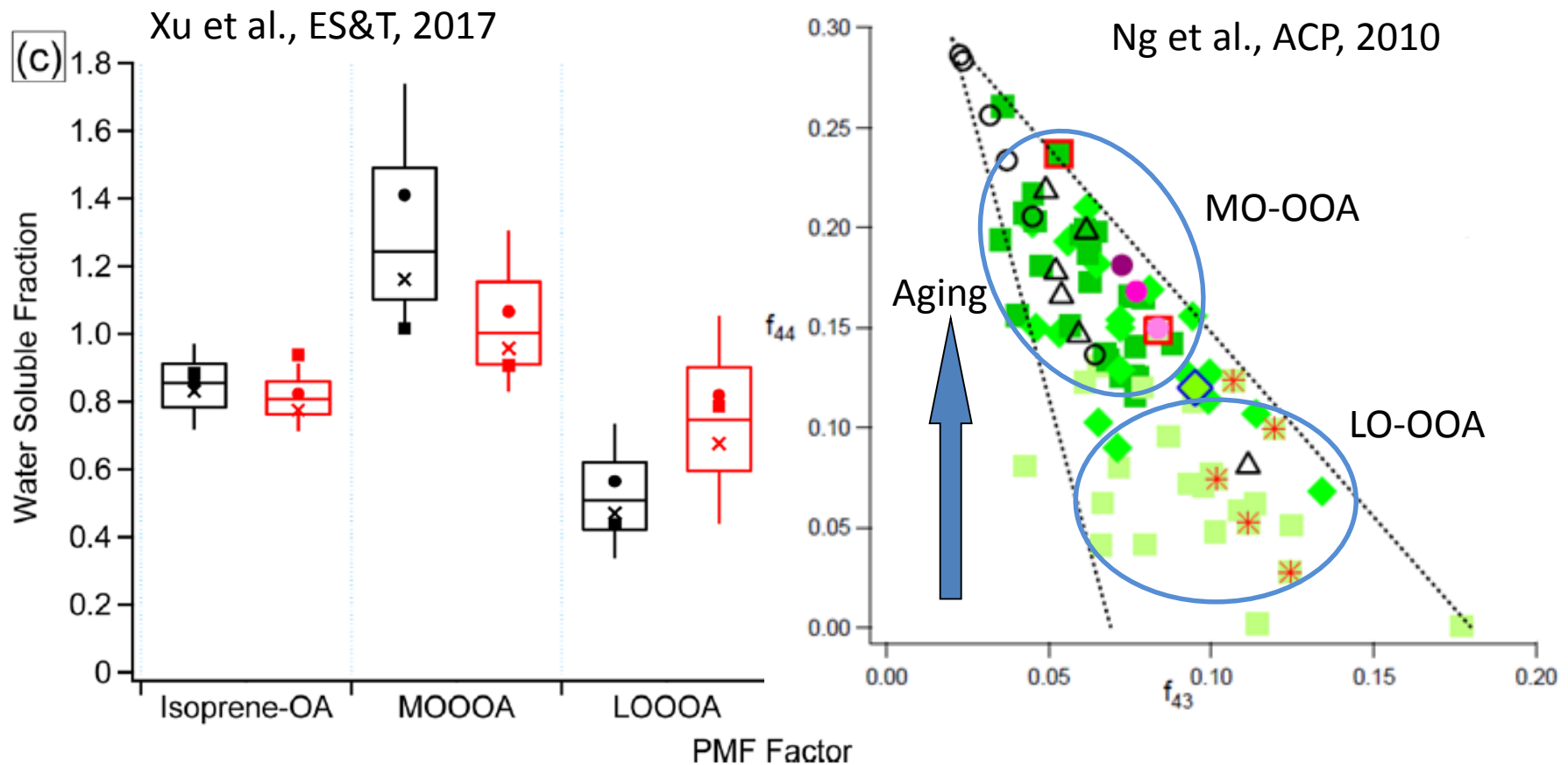


- Isoprene-OA is water soluble.  
→ Aqueous-phase reaction.
- MOOOA is water soluble.  
→ Highly oxidized species.
- 40-80% of LOOOA is water soluble  
→ Contribution from organic nitrates (largely water-insoluble).
- Half of BBOA is water soluble.
- COA and HOA are largely water insoluble.

Xu et al., ES&T, 2017

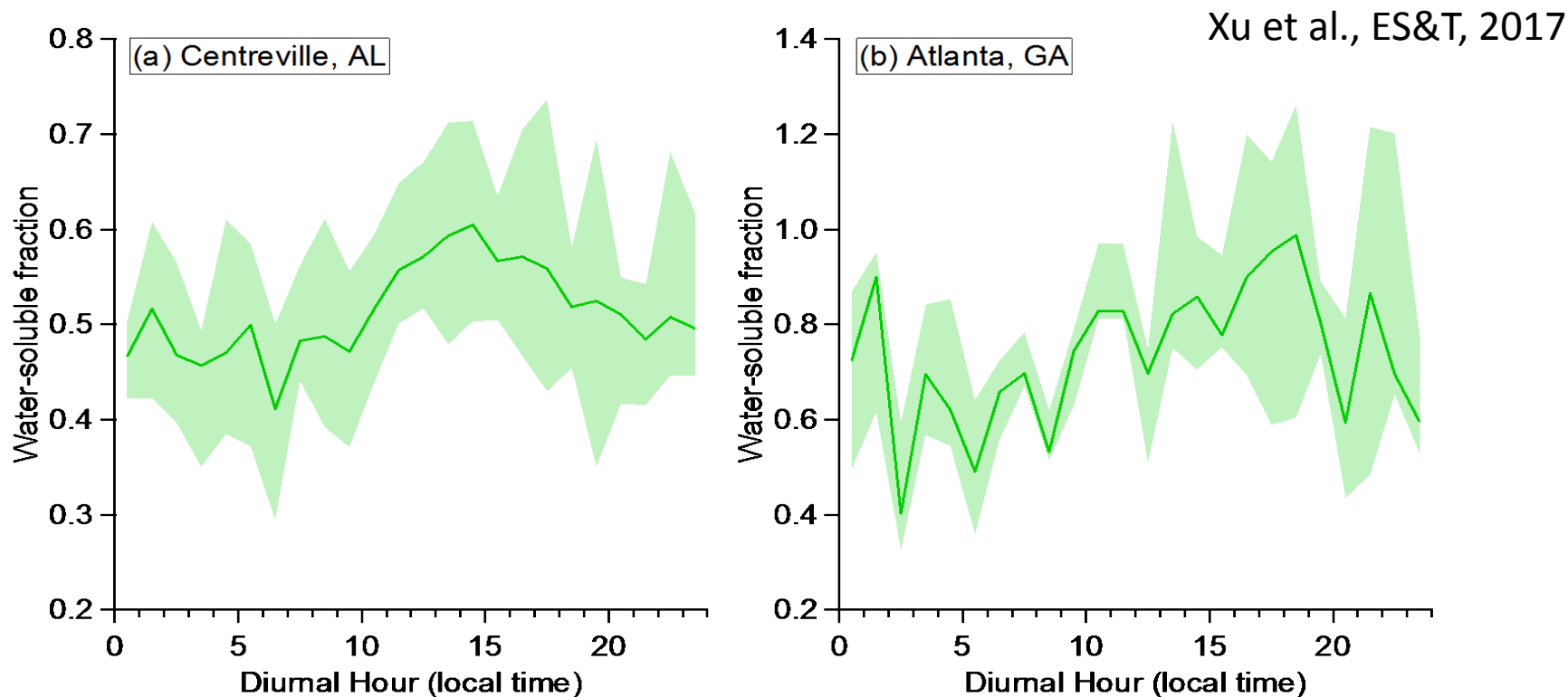


# Water-Solubility of OA Factors



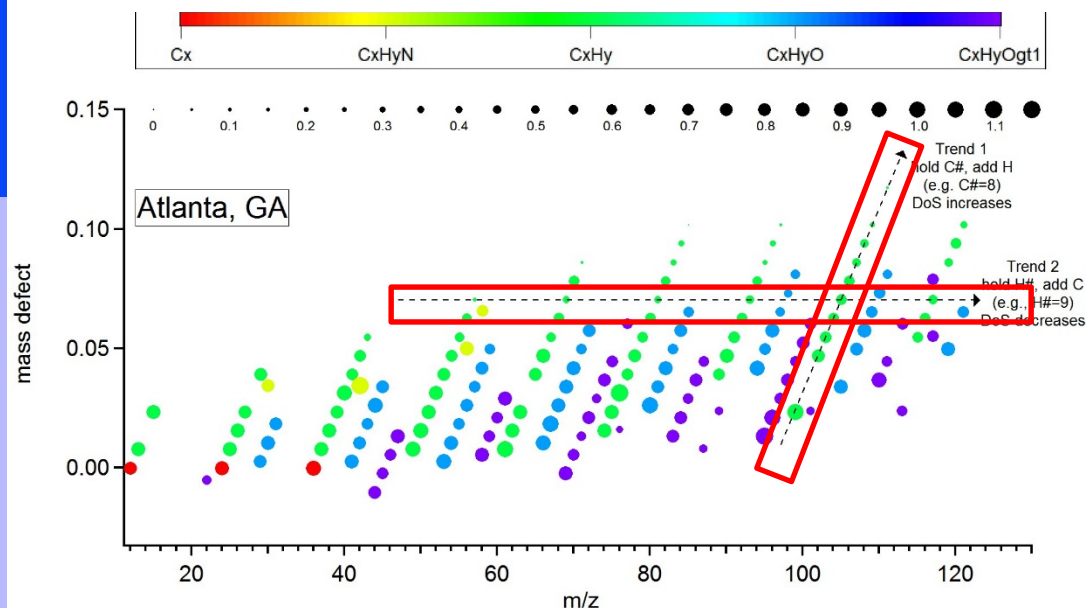
- Solubility of MO-OOA > LO-OOA; aerosols become more oxidized and hygroscopic with atmospheric aging
- MO-OOA: wide distribution suggests a variety of sources

# Diurnal Trends of LO-OOA Water-Solubility



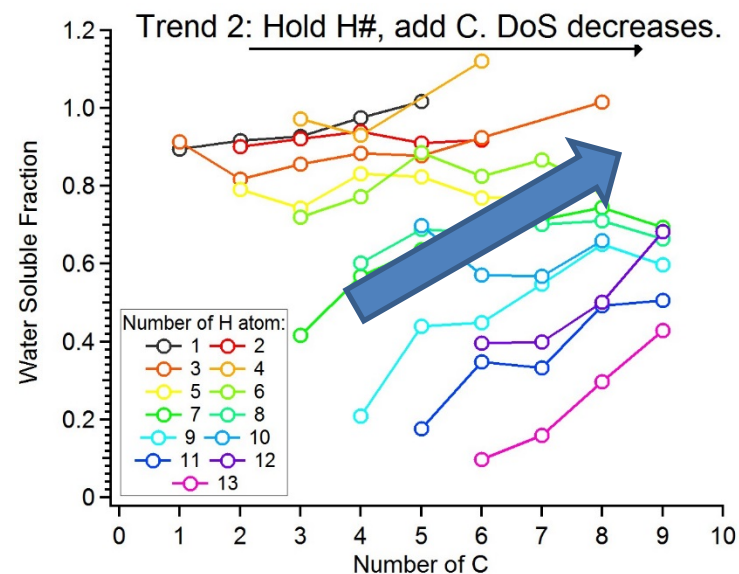
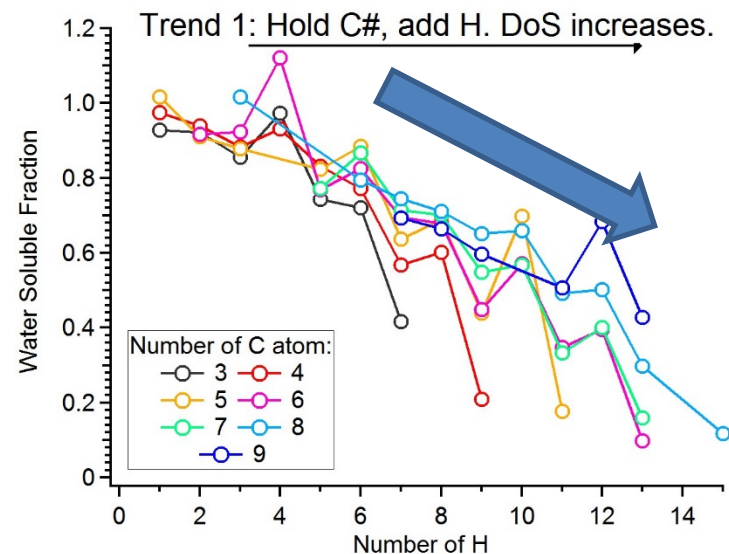
- Organic nitrates from monoterpene +  $\text{NO}_3$  contributes to LO-OOA
- Organic nitrates fairly water insoluble (e.g., Suda et al., 2014; Pye et al., 2016)
- Smaller fraction of LO-OOA is water-soluble at night than in the day, which is consistent with more abundant organic nitrates at night.

# Water-Solubility of AMS Ions



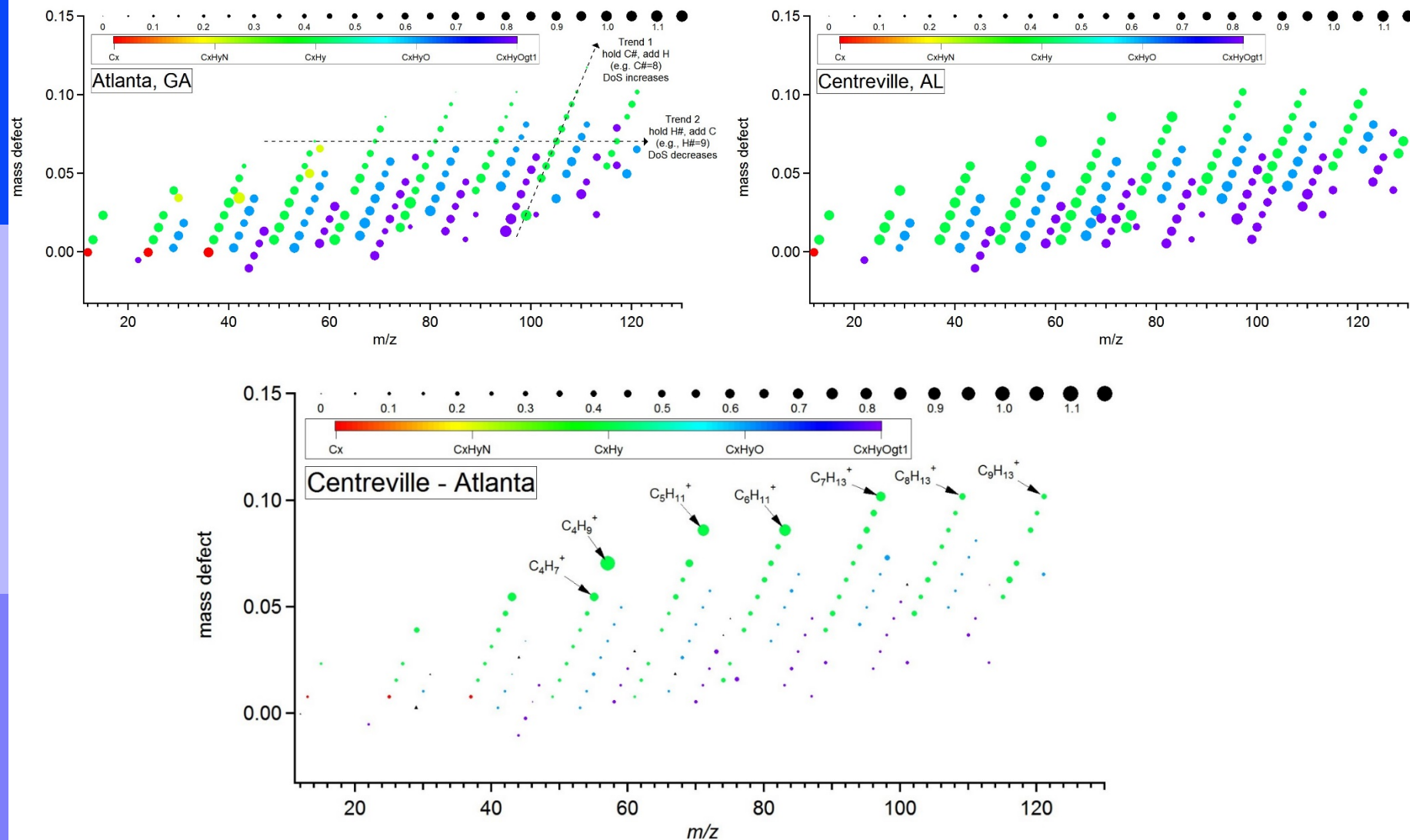
Higher degree of saturation of the  $C_xH_y^+$  ions  
 → Lower water solubility

Xu et al., ES&T, 2017



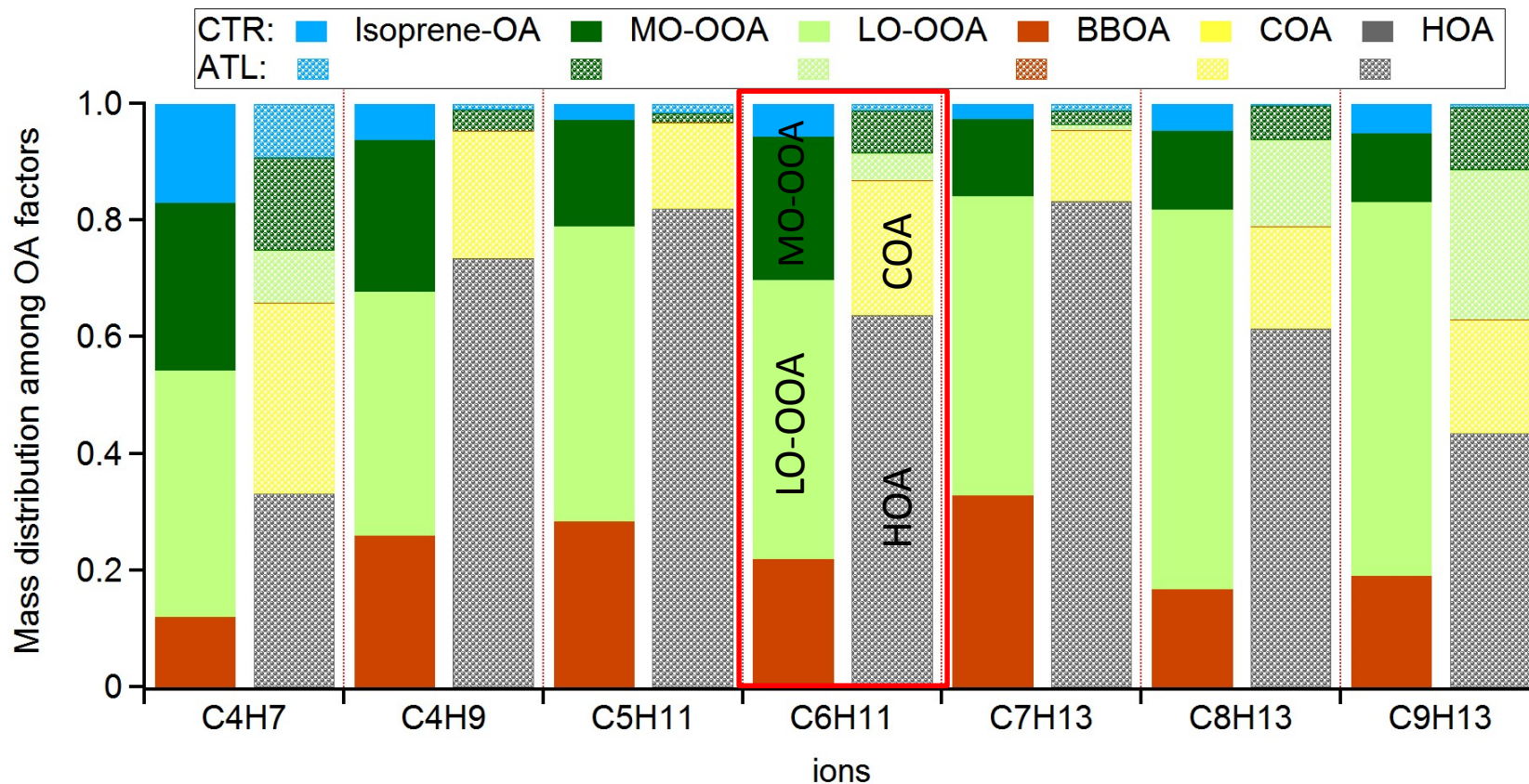
# Water-Solubility of AMS Ions

Xu et al., ES&T, 2017



- The water-solubility of C<sub>x</sub>H<sub>y</sub><sup>+</sup> ions with higher degree of saturation (i.e., C<sub>4</sub>H<sub>7</sub><sup>+</sup>, C<sub>4</sub>H<sub>9</sub><sup>+</sup>, C<sub>5</sub>H<sub>11</sub><sup>+</sup>, C<sub>6</sub>H<sub>11</sub><sup>+</sup>, C<sub>7</sub>H<sub>13</sub><sup>+</sup>, C<sub>8</sub>H<sub>13</sub><sup>+</sup>, and C<sub>9</sub>H<sub>13</sub><sup>+</sup>) is substantially larger in CTR than ATL.

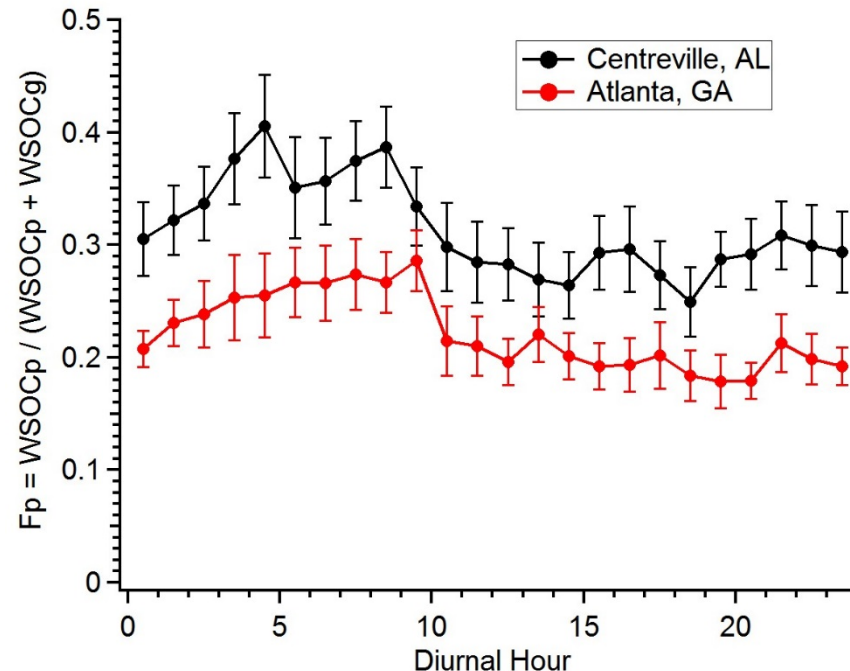
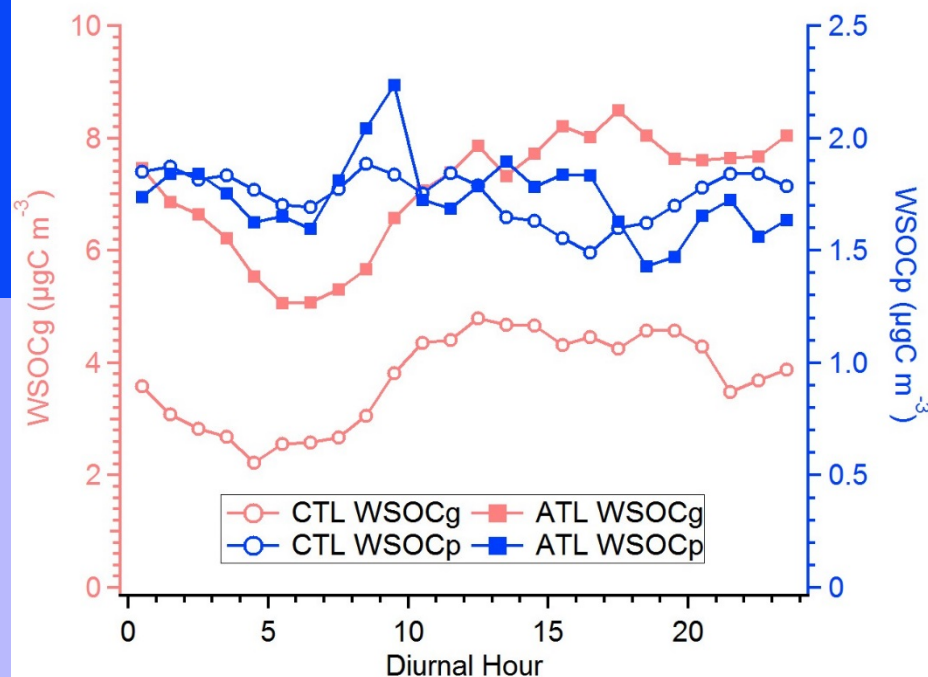
# Water-Solubility of AMS Ions



- These  $C_xH_y^+$  ions have different sources at the two sites.
- Atlanta: HOA and COA → long-chain alkanes from vehicle emissions or fatty acids from cooking.
- Centreville: MO-OOA and LO-OOA → secondary production from monoterpenes and sesquiterpenes etc.



# WSOC Gas/Particle Partitioning



- The  $\text{WSOC}_g$  concentration is two and four times higher than that of  $\text{WSOC}_p$ .
- $\text{WSOC}_g$  peaks in the afternoon.  $\text{WSOC}_p$  is relatively constant.
- $F_p$  is lower in Atlanta than Centreville. This is likely because of higher concentrations of small organic acids and carbonyls in Atlanta, which arise from the fragmentation during the oxidation of VOCs under high- $\text{NO}_x$  conditions

# Implications for Offline AMS Analysis of Filter Samples

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- Water soluble extracts; source apportionment of WSOM extrapolated to ambient OA using a recovery ratio
- Accuracy of ambient OA source apportionment depends on the recovery ratio of the water-solubility of OA factors
- Things to keep in mind / Uncertainties; More work needed
  - Recovery ratio is a surrogate for, but not identical to, water-solubility
  - Water-solubility of OA factors varies with time and location; recovery ratio derived from one study may not be applicable to another study at a different site
  - Challenges in estimating ambient concentration of HOA or OA from any sources with low water-solubility

# Conclusions

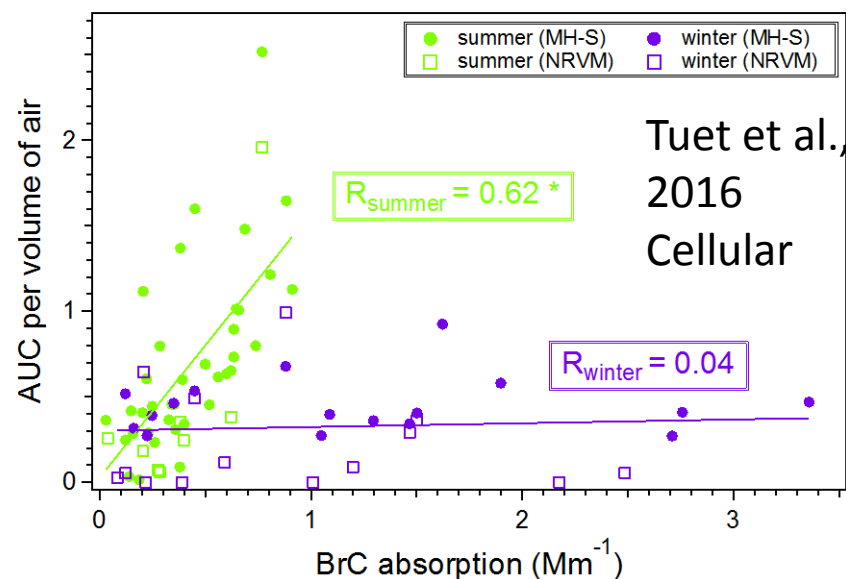
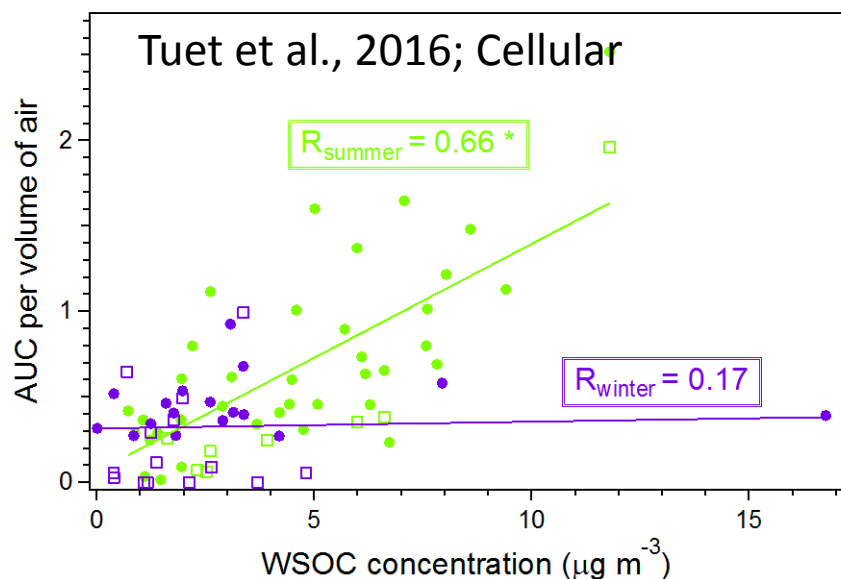
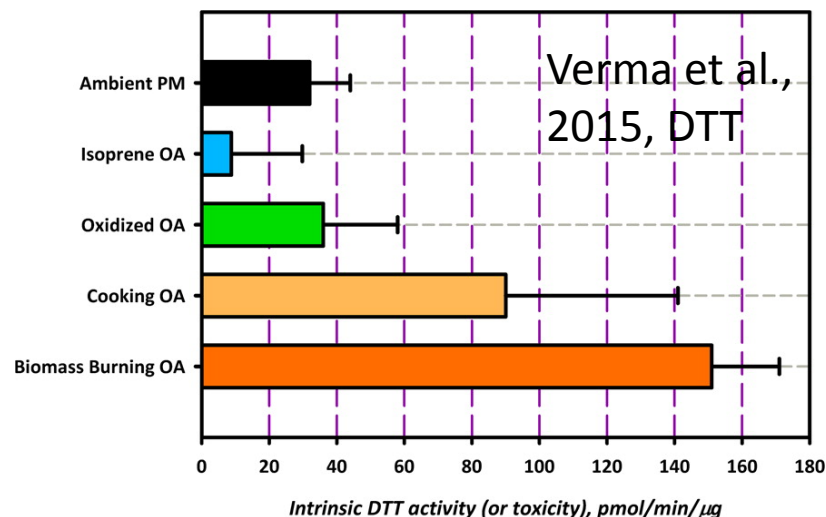
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Approximately 88% and 77% of OA are water-soluble in rural Centreville and urban Atlanta, respectively.

PMF factor	Water-solubility	Interpretation
Isoprene-OA	>80%	represents OA formed through the aqueous-phase reaction of isoprene epoxydiols
MO-OOA	~100%	represents highly oxidized compounds
LO-OOA	47-68%	includes contributions from organic nitrates
BBOA and COA	a wide range	diverse burning conditions and cooking activities
HOA	~0%	primary emissions from combustions

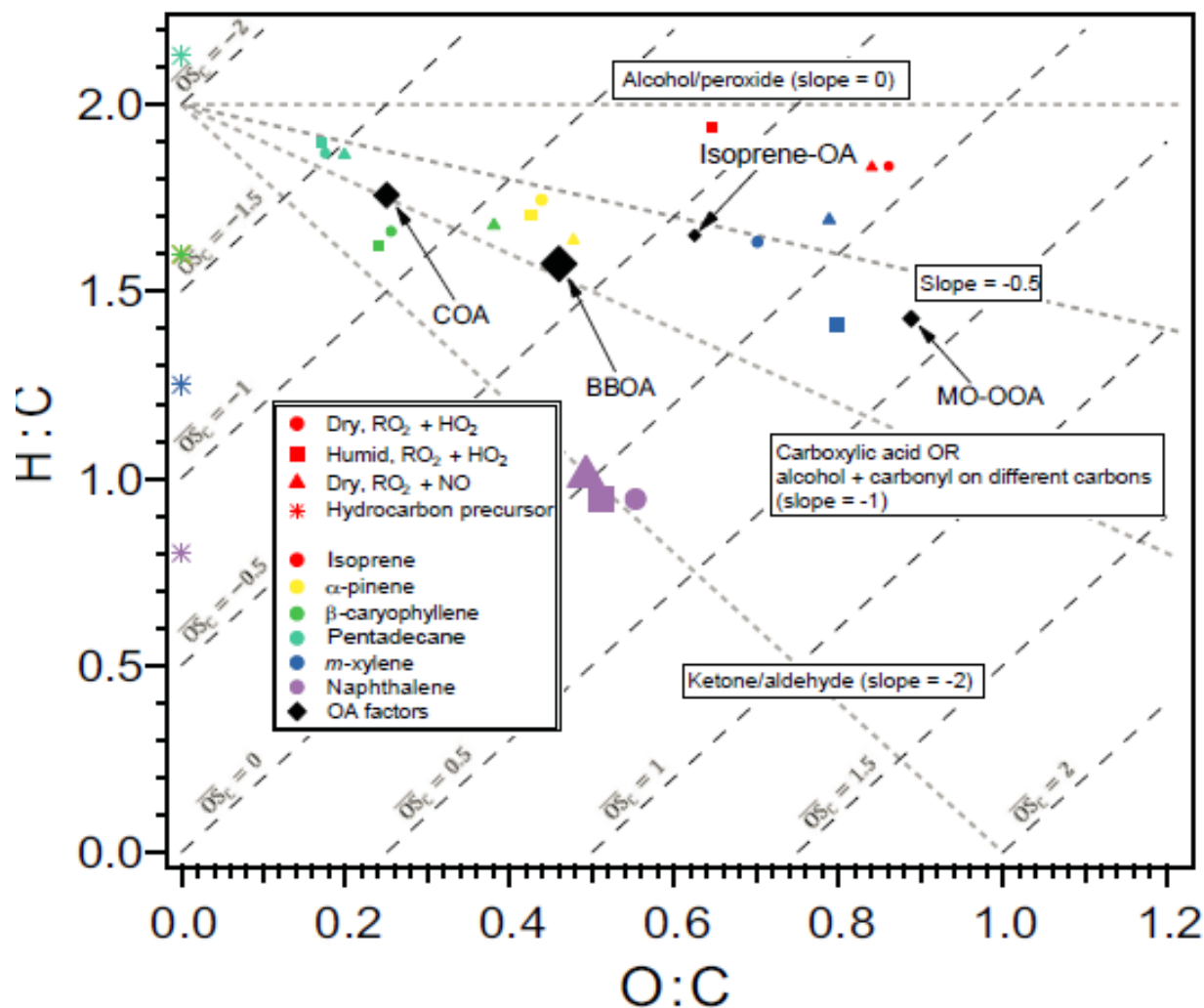
# Health Effects of Aerosols

- DTT Assay (chemical assay, oxidative potential)
- Macrophage assay (cellular assay, reaction oxygen species, ROS, generation)
- Linking aerosol composition to health effects



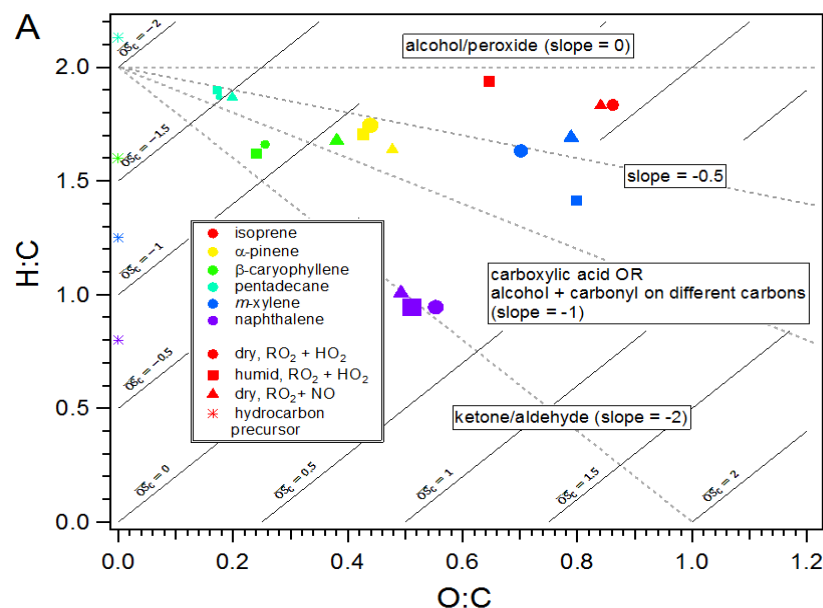
# DTT Activity of Different Types of SOA (Chamber Data)

Size of data point = DTT activity, see details in Tuet et al., ACP, 2017

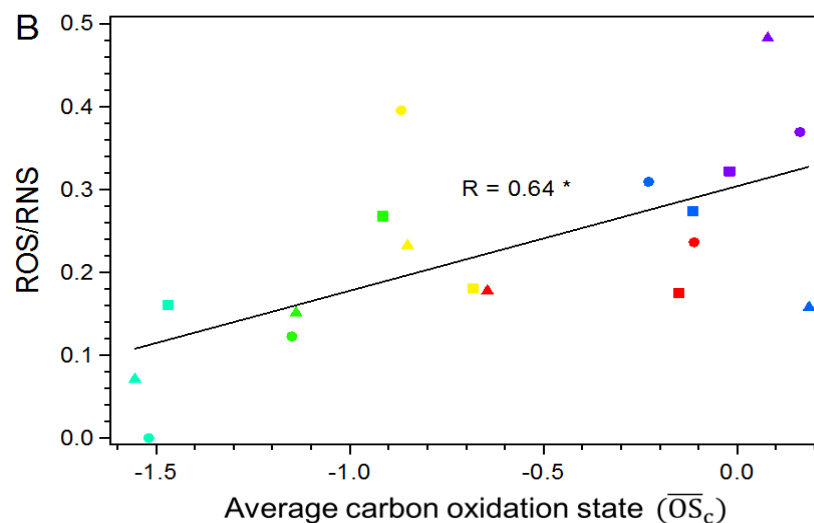




# Cellular ROS of Different Types of SOA (Chamber Data)



Size of data point = cellular reactive oxygen species (ROS), see details in Tuet et al., ACPD, 2017



# References

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## Water Soluble Organics:

See Supplementary Information of publications for PMF details  
Users responsibility!

Xu, L., Guo, H., Weber, R. J., and Ng, N. L.: Chemical Characterization of Water-Soluble Organic Aerosol in Contrasting Rural and Urban Environments in the Southeastern United States, *Environ. Sci. Technol.*, 51, 78-88, 2017.

## Aerosol Health Effects:

Tuet, W. Y., Fok, S., Verma, V., Rodriguez, M. S. T., Grosberg, A., Champion, J. A., and Ng, N. L.: Dose-dependent intracellular reactive oxygen and nitrogen species (ROS/RNS) production from particulate matter exposure: comparison to oxidative potential and chemical composition, *Atmos. Environ.*, 144, 335-344, 2016.

Tuet, W. Y., Chen, Y., Xu, L., Fok, S., Gao, D., Weber, R. J., and Ng, N. L.: Chemical oxidative potential of secondary organic aerosol (SOA) generated from the photooxidation of biogenic and anthropogenic volatile organic compounds, *Atmos. Chem. Phys.*, 17, 839-853, doi:10.5194/acp-17-839-2017, 2017.

Tuet, W. Y., Chen, Y., Fok, S., Champion, J. A., and Ng, N. L.: Inflammatory responses to secondary organic aerosols (SOA) generated from biogenic and anthropogenic precursors, *Atmos. Chem. Phys. Discuss.*, doi:10.5194/acp-2017-262, 2017.

Verma, V., Fang, T., Xu, L., Peltier, R. E., Russell, A. G., Ng, N. L., and Weber, R. J.: Organic Aerosols Associated with the Generation of Reactive Oxygen Species (ROS) by Water-Soluble PM<sub>2.5</sub>, *Environ. Sci. Technol.*, 49, 4646-4656, 10.1021/es505577w, 2015.

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