

**AMS Introduction**

**Doug Worsnop**

**AMS Users Meeting**

~~Aerodyne~~

~~Caltech~~

~~Georgia Tech~~

~~FZ - Juelich~~

~~University of Minnesota~~

**Desert Research Institute**

~~October 2001/2002~~

~~October 24, 2003~~

~~October 8, 2004~~

~~25 August, 2005~~

~~16 September, 2006~~

**29 September, 2007**

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~~October 13, 2002~~

~~October 24, 2003~~

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~~25 August, 2005~~

~~16 September, 2006~~

**29 September, 2007**

**Nitrate**

**Sulfate**

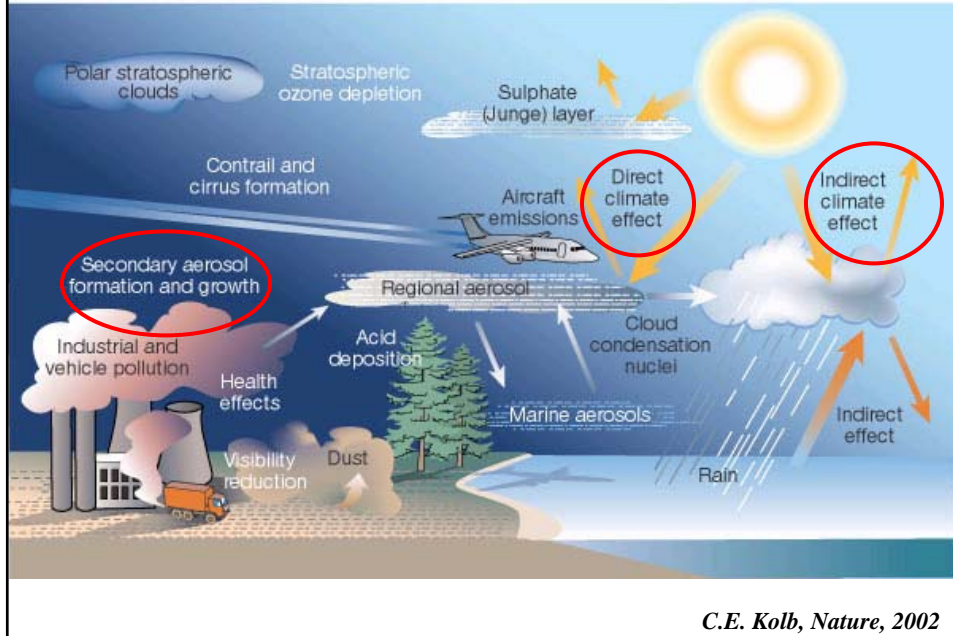
**Ammonia**

**Organics (44,57)**

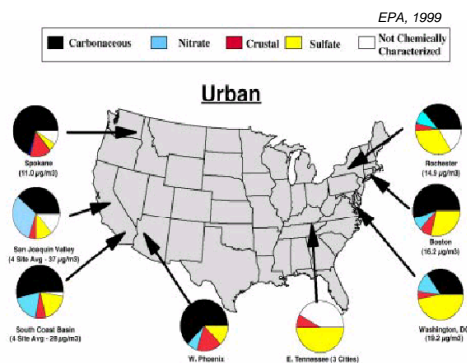
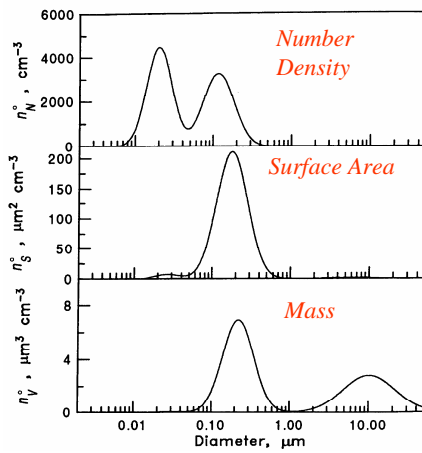
**OOA, HOA**

**OOA1, OOA2**

# Aerosols in the Atmosphere



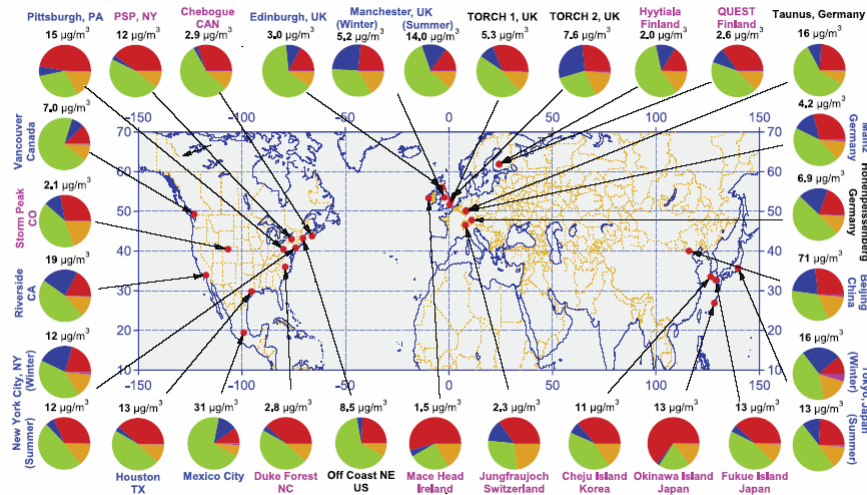
## Ambient Aerosol Size Distribution



**Goal: Size-Resolved Chemical Composition of Ambient Aerosol**

**Ubiquity and dominance of oxygenated species in organic aerosols in anthropogenically-influenced Northern Hemisphere midlatitudes**

Q. Zhang,<sup>1</sup> J. L. Jimenez,<sup>2</sup> M. R. Canagaratna,<sup>3</sup> J. D. Allan,<sup>4</sup> H. Coe,<sup>4</sup> I. Ulbrich,<sup>2</sup> M. R. Alfarra,<sup>5</sup> A. Takami,<sup>6</sup> A. M. Middlebrook,<sup>7</sup> Y. L. Sun,<sup>1</sup> K. Dzepina,<sup>2</sup> E. Dunlea,<sup>2</sup> K. Docherty,<sup>2</sup> P. F. DeCarlo,<sup>2</sup> D. Salcedo,<sup>8</sup> T. Onasch,<sup>2</sup> J. T. Jayne,<sup>2</sup> T. Miyoshi,<sup>6</sup> A. Shimono,<sup>9</sup> S. Hatakeyama,<sup>9</sup> N. Takegawa,<sup>10</sup> Y. Kondo,<sup>10</sup> J. Schneider,<sup>10</sup> F. Drewnick,<sup>11</sup> S. Borrmann,<sup>11</sup> S. Weimer,<sup>1</sup> K. Demerjian,<sup>1</sup> P. Williams,<sup>4</sup> K. Bower,<sup>4</sup> R. Bahreini,<sup>2,7</sup> L. Cottrell,<sup>12</sup> R. J. Griffin,<sup>12</sup> J. Rautiaainen,<sup>13</sup> J. Y. Sun,<sup>14</sup> Y. M. Zhang,<sup>14</sup> and D. R. Worsnop<sup>3</sup>



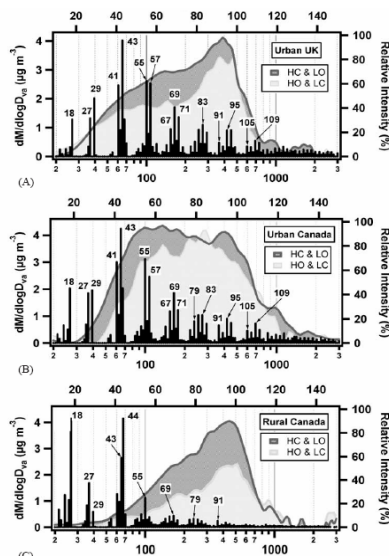
GEOPHYSICAL RESEARCH LETTERS, VOL. 34, L13801, doi:10.1029/2007GL029979, 2007

**ATMOSPHERIC ENVIRONMENT**

**Characterization of urban and rural organic particulate in the Lower Fraser Valley using two Aerodyne Aerosol Mass Spectrometers**

138 (2004) 5745–5758

M. Rami Alfarra<sup>a,\*</sup>, Hugh Coe<sup>b</sup>, James D. Allan<sup>b</sup>, Keith N. Bower<sup>b</sup>, Hacene Boudries<sup>c</sup>, Manjula R. Canagaratna<sup>a</sup>, Jose L. Jimenez<sup>d</sup>, John T. Jayne<sup>e</sup>, Arthur A. Garforth<sup>a</sup>, Shao-Meng Li<sup>c</sup>, Douglas R. Worsnop<sup>c</sup>



**Manchester center**

*Small particles <200nm*

*Hydrocarbon organics*

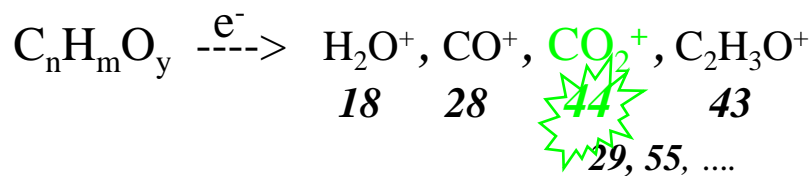
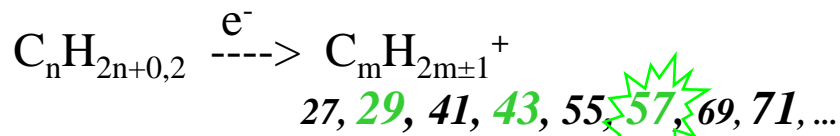
**Vancouver center**

**Rural (outside Vancouver)**

*Larger particles (>200 nm)*

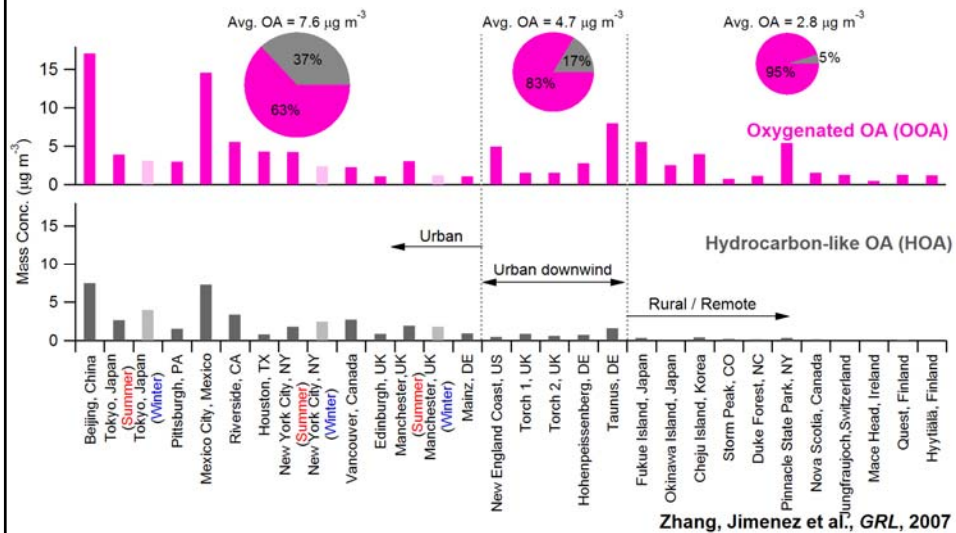
*m/z 44 CO<sub>2</sub><sup>+</sup> dominates*

## Organic Mass Spectra



*Following flash vaporization at ~600°C*

## Organic Aerosols: Urban vs. Rural/Remote

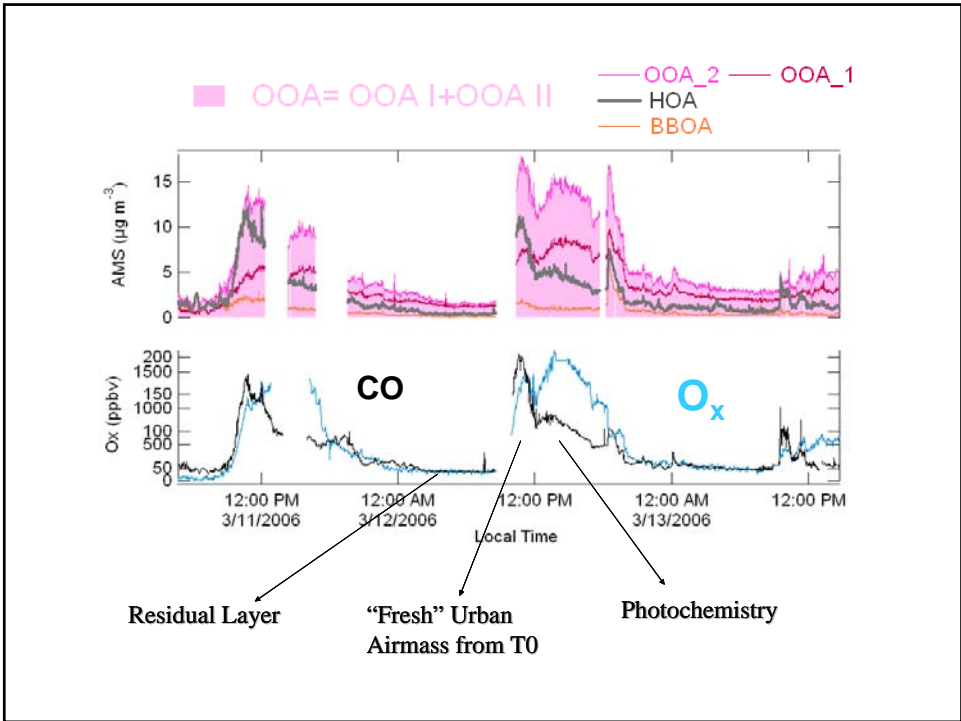


### AMS (PMF) Factors

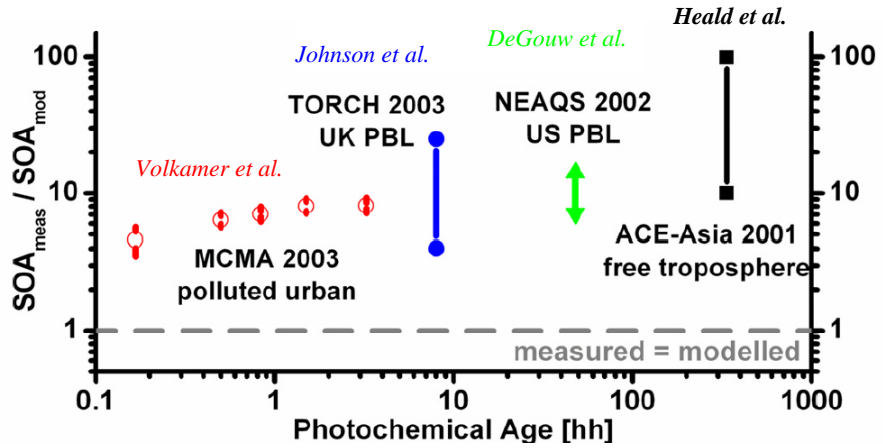
	600C, e-	m/z
HOA:	$C_nH_m \rightarrow C_{n-x}H_{m-y}^+$	27,29,41,43,55,57,69,71,...
<i>LESS OXIDIZED</i>		
OOA2	$C_nH_mO \rightarrow C_2H_3O^+, C_3H_3O^+, R^+$	43,55, ...
<i>MORE OXIDIZED</i>		
OOA1	$C_nH_mO_2 \rightarrow CO_2^+, HCO_2^+, R^+$	44,45, ...
BBOA	$R \rightarrow R^+, C_2H_4O_2^+, C_3H_3O_2^+$	60,73, .... <i>levoglucosan</i>

**O:C ratio:** HOA << OOA2 ~ BBOA < OOA1

*Factors are not unique or identical among campaigns, platforms*  
**CONSISTENT TRENDS**



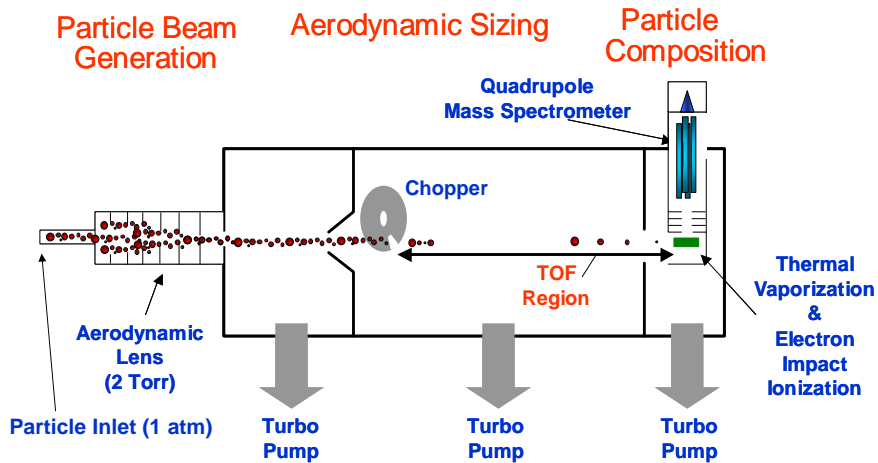
# SOA: measurements vs. models



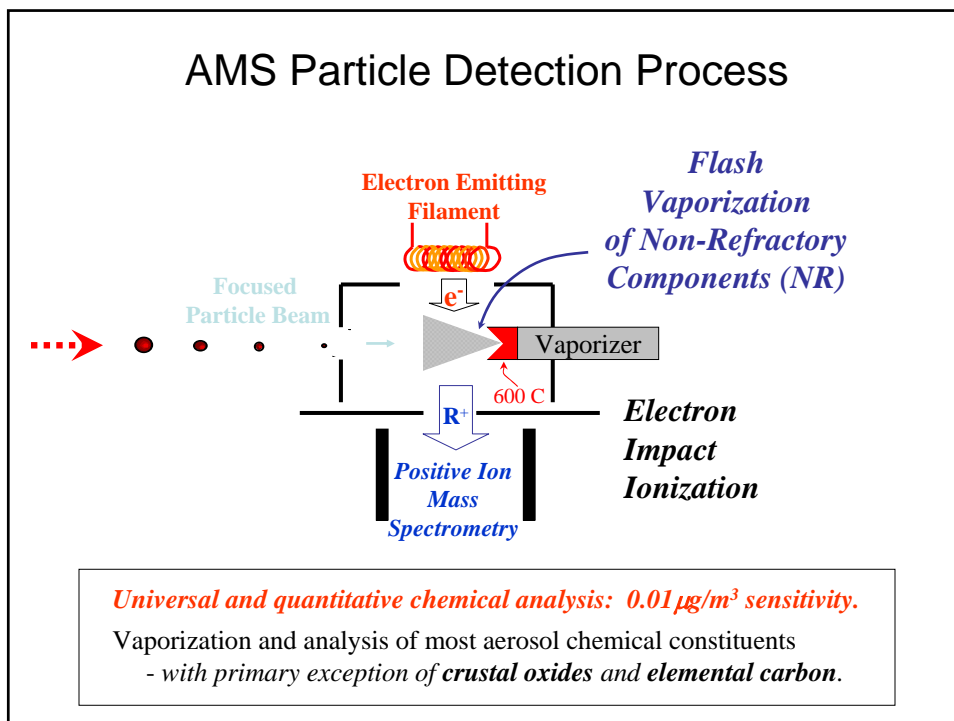
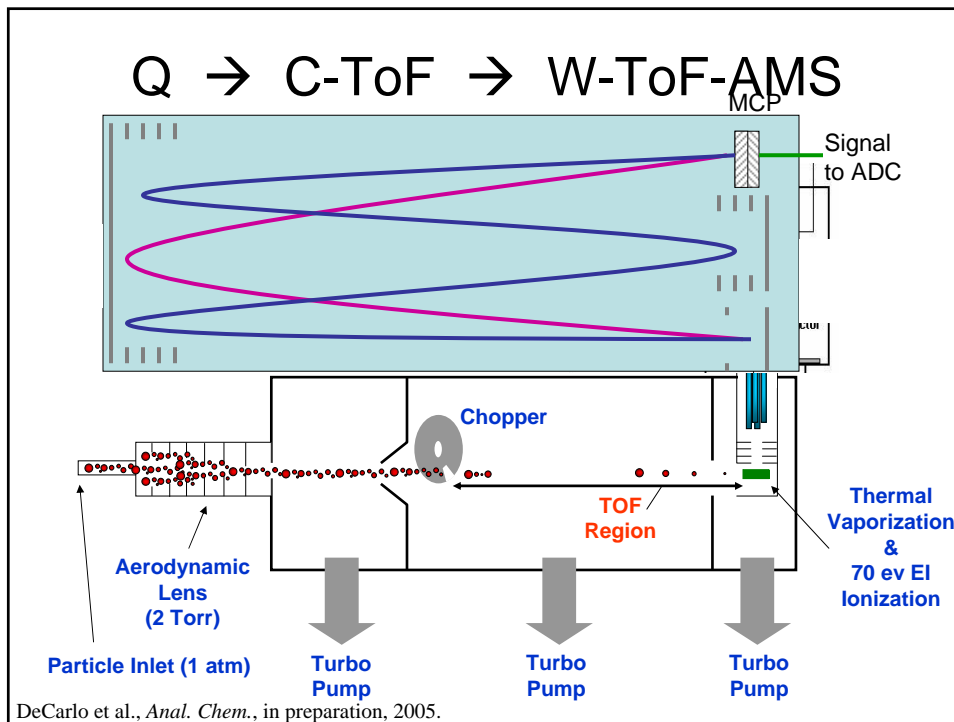
- **SOA correlates with photochemistry:  $O_x$**

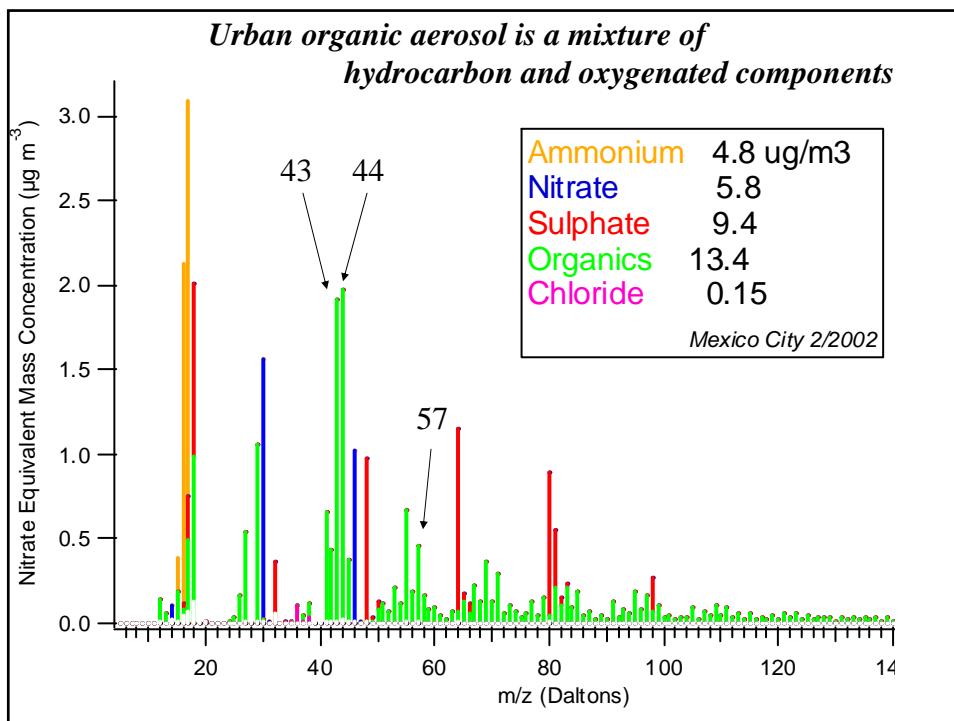
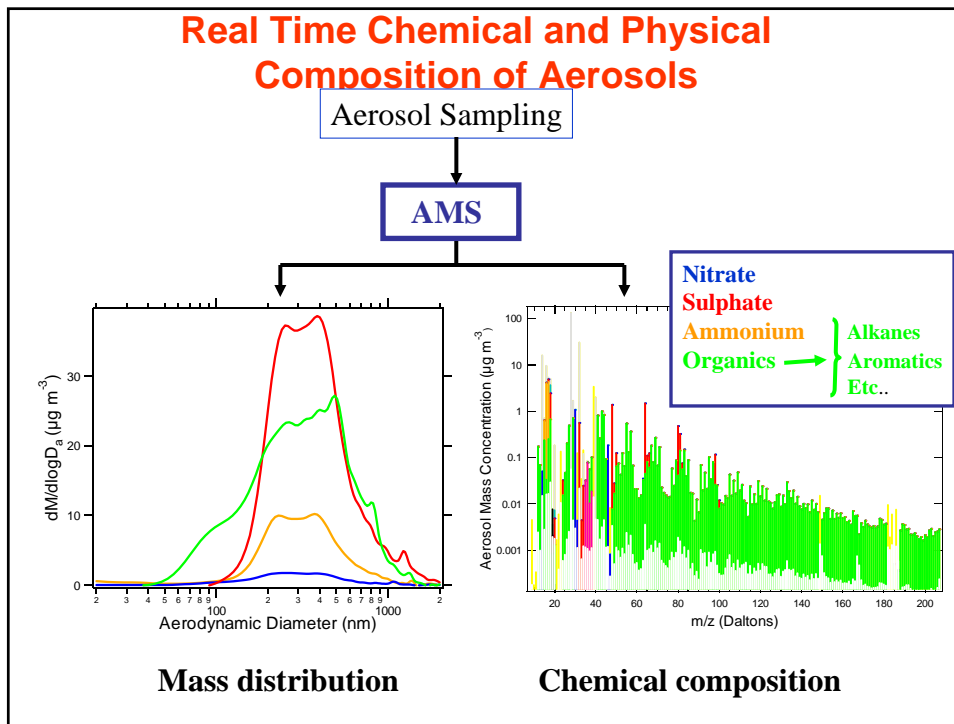
*Volkamer, Jimenez, et al GRL, 2006*

# Aerosol Mass Spectrometer (AMS)



- Efficient transmission (40-1000 nm), aerodynamic sizing, linear mass signal
- Non-refractory PM1.0 mass loadings and chemically-specified mass distributions







**Same AMS, same issues, new issues**

“CE” collection efficiency  
**particle bounce**      **CE ~ 0.5, except ...**

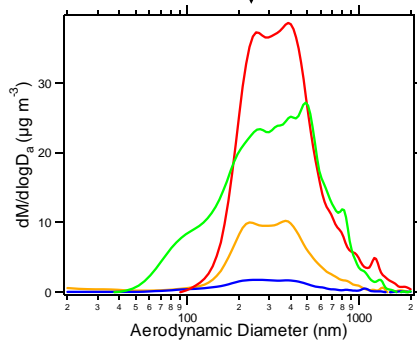
“RIE” relative ionization efficiency  
- particularly **RIE<sub>org</sub> ~ 1.4**

“ToF threshold & baseline” - single ion determination  
linearity of large and small signals (**dynamic range**)

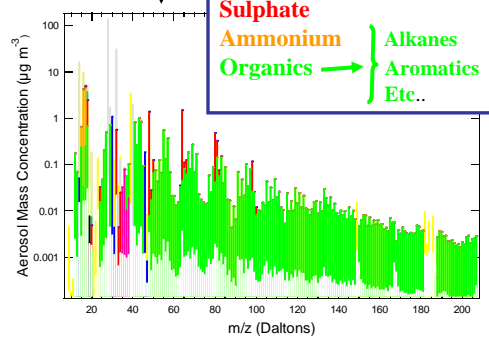
**Real Time Chemical and Physical  
Composition of Aerosols**

Aerosol Sampling

AMS



**Mass distribution**



**Chemical composition**

## MS Signatures for Aerosol Species Identification

*color coded to match spectra*

Group	Molecule/Species	Ion Fragments	Mass Fragments
Water	H <sub>2</sub> O	$e^- \rightarrow$ H <sub>2</sub> O <sup>+</sup> , HO <sup>+</sup> , O <sup>+</sup>	18, 17, 16
Ammonium	NH <sub>3</sub>	$e^- \rightarrow$ NH <sub>3</sub> <sup>+</sup> , NH <sub>2</sub> <sup>+</sup> , NH <sup>+</sup>	17, 16, 15
Nitrate	HNO <sub>3</sub>	$e^- \rightarrow$ HNO <sub>3</sub> <sup>+</sup> , NO <sub>2</sub> <sup>+</sup> , NO <sup>+</sup>	63, 46, 30
Sulfate	H <sub>2</sub> SO <sub>4</sub>	$e^- \rightarrow$ H <sub>2</sub> SO <sub>4</sub> <sup>+</sup> , HSO <sub>3</sub> <sup>+</sup> , SO <sub>3</sub> <sup>+</sup> SO <sub>2</sub> <sup>+</sup> , SO <sup>+</sup>	98, 81, 80 64, 48
Organic (Oxygenated)	C <sub>n</sub> H <sub>m</sub> O <sub>y</sub>	$e^- \rightarrow$ H <sub>2</sub> O <sup>+</sup> , CO <sup>+</sup> , CO <sub>2</sub> <sup>+</sup> H <sub>3</sub> C <sub>2</sub> O <sup>+</sup> , HCO <sub>2</sub> <sup>+</sup> , C <sub>n</sub> H <sub>m</sub> <sup>+</sup>	18, 28, 44 43, 45, ...
Organic (hydrocarbon)	C <sub>n</sub> H <sub>m</sub>	$e^- \rightarrow$ C <sub>n</sub> H <sub>m</sub> <sup>+</sup>	27,29,41,43,55,57,69,71...

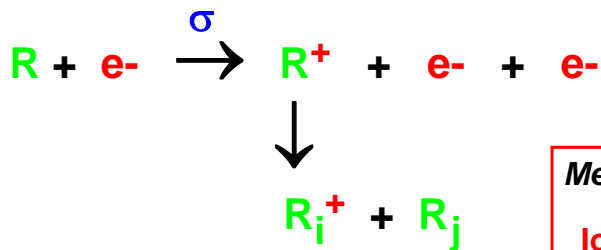
*Standard electron impact ionization @ 70 eV*

*Easy to quantify: ca. NIST MS library*

*Easy to separate inorganic and organic components*

*Speciation of organic composition is challenging*

### ELECTRON IMPACT (EI) IONIZATION



**Measure all ions:**

$$\text{Ion Rate} = \sum R_i^+$$

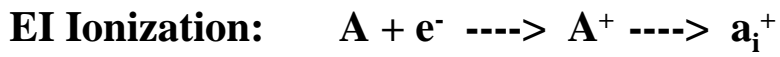
EI Cross Section ( $\sigma$ )  $\propto$  **electrons/molecule**

$\propto$  **mass/molecule**

**Ion Rate** = **2<sup>ndary</sup> electrons/sec**  $\propto$   $\sigma \cdot$  **molecules/sec**

**Ion Rate** =  $\sum R_i^+$   $\propto$  **total mass/sec**

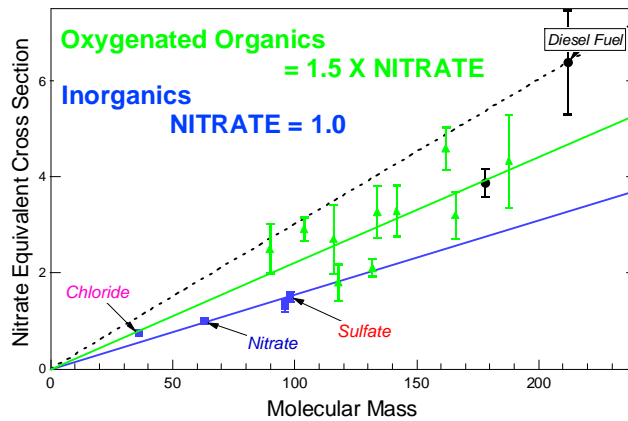
*independent of parent or fragment (neutral or ion)  
molecular mass*



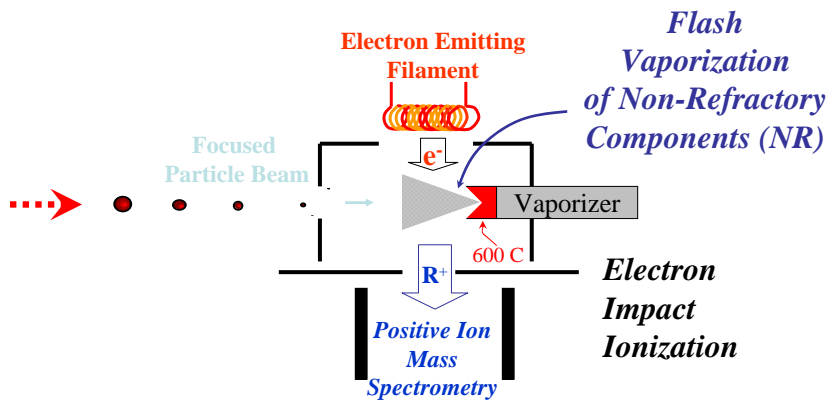
**Mass Loading A**  $\propto (MW_A/IE_A) \Sigma \text{ Ion Signal}$

**Calibration Factor \*  $(MW_{NO_3}/IE_{NO_3})$**

**EI Ionization  
Cross  
Sections**



## AMS Particle Detection



*Universal / quantitative chemical analysis:  $0.01 \mu\text{g}/\text{m}^3$  sensitivity.*

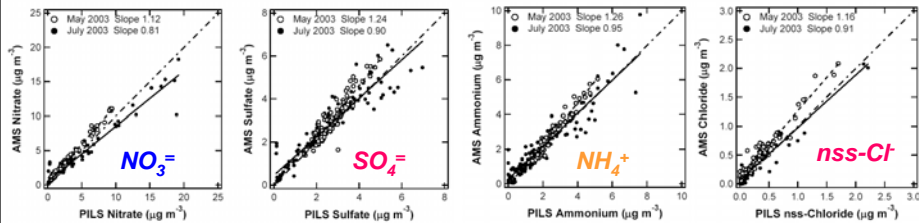
**“EVERYTHING AT THE SAME TIME”**

**No Separation - bulk analysis**

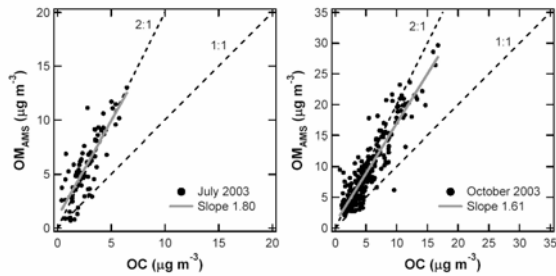
*[ no dust (oxides) or elemental (black) carbon ]*

## Characterization of an Aerodyne Aerosol Mass Spectrometer (AMS): Intercomparison with Other Aerosol Instruments

N. Takegawa,<sup>1</sup> Y. Miyazaki,<sup>1</sup> Y. Kondo,<sup>1</sup> Y. Komazaki,<sup>1</sup> T. Miyakawa,<sup>1</sup> J. L. Jimenez,<sup>2</sup> J. T. Jayne,<sup>3</sup> D. R. Worsnop,<sup>3</sup> J. D. Allan,<sup>4</sup> and R. J. Weber<sup>5</sup>



OM vs OC

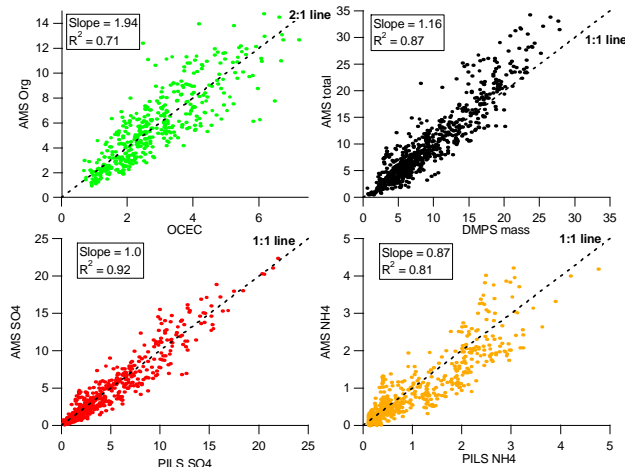


Takegawa et al., 2005

Assumed CE = 0.5

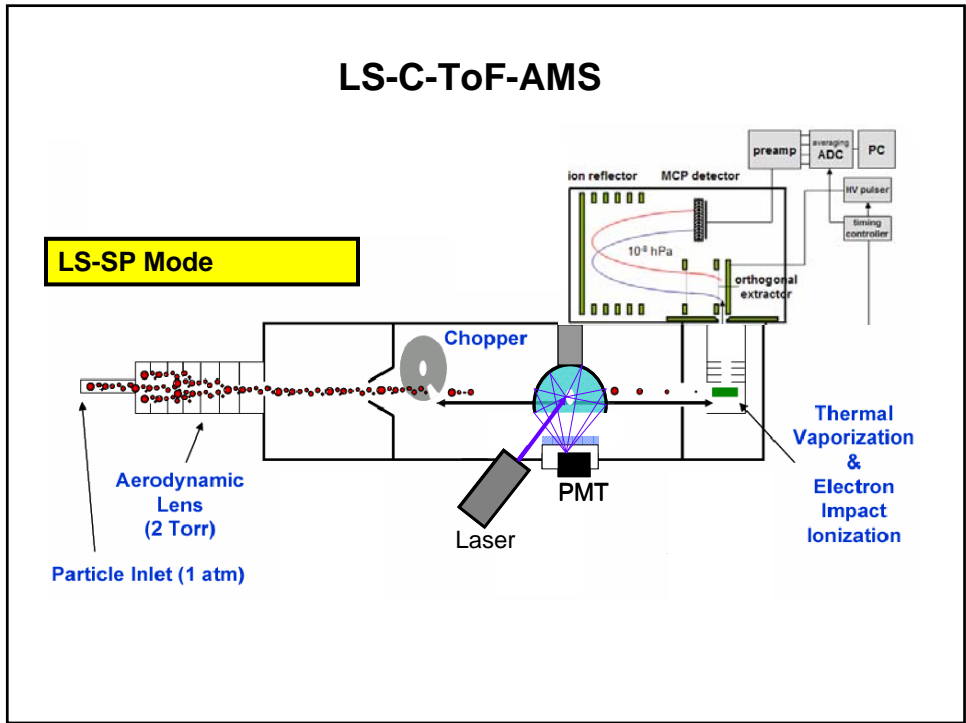
$$\frac{\text{OM}}{\text{OC}} = \frac{C_n H_m O_p}{C_n}$$

## AMS (NR-PM<sub>1</sub>) quantification

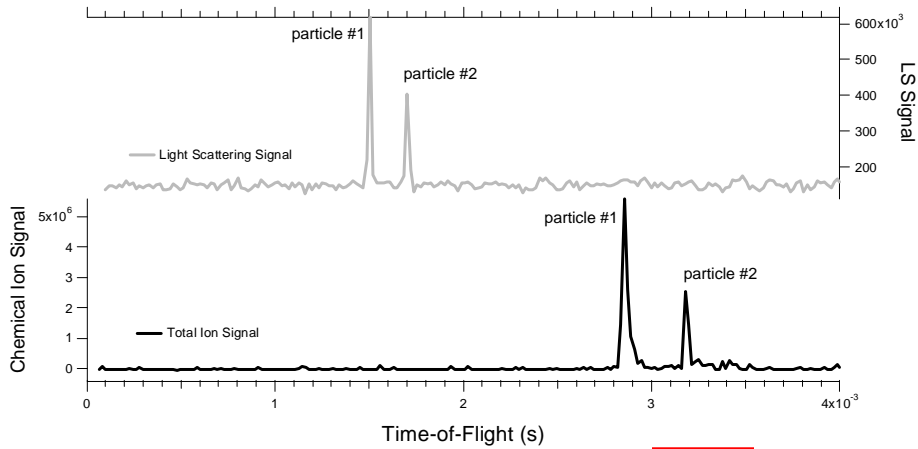


- Composition/phase-dependent collection efficiency (CE) applied
- Correction for smaller cut diameter in AMS lens compared to PM<sub>1</sub> impactors

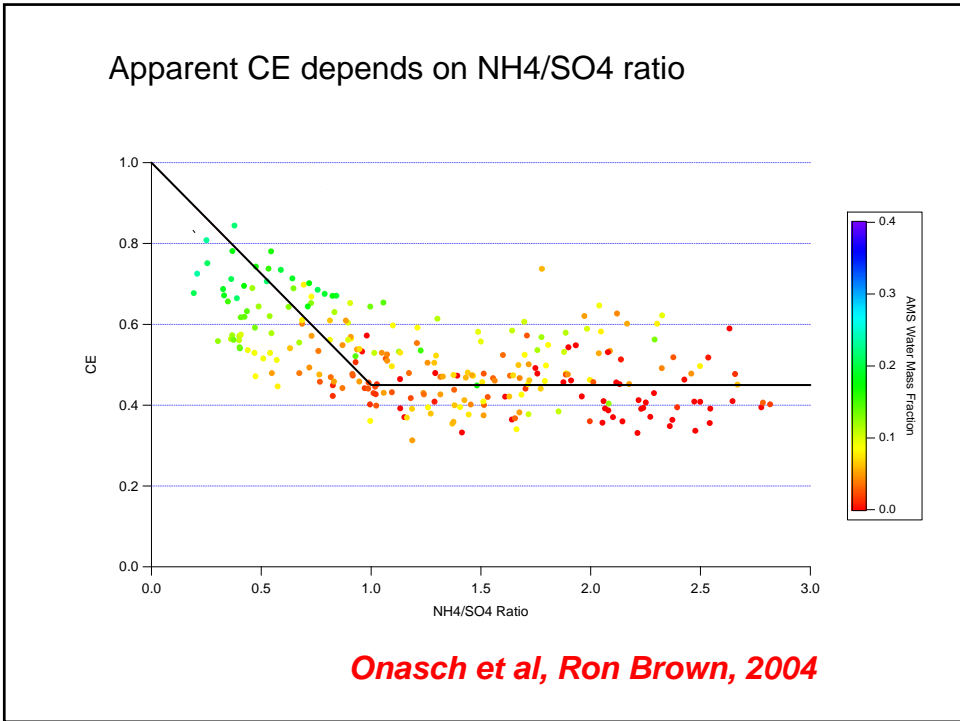
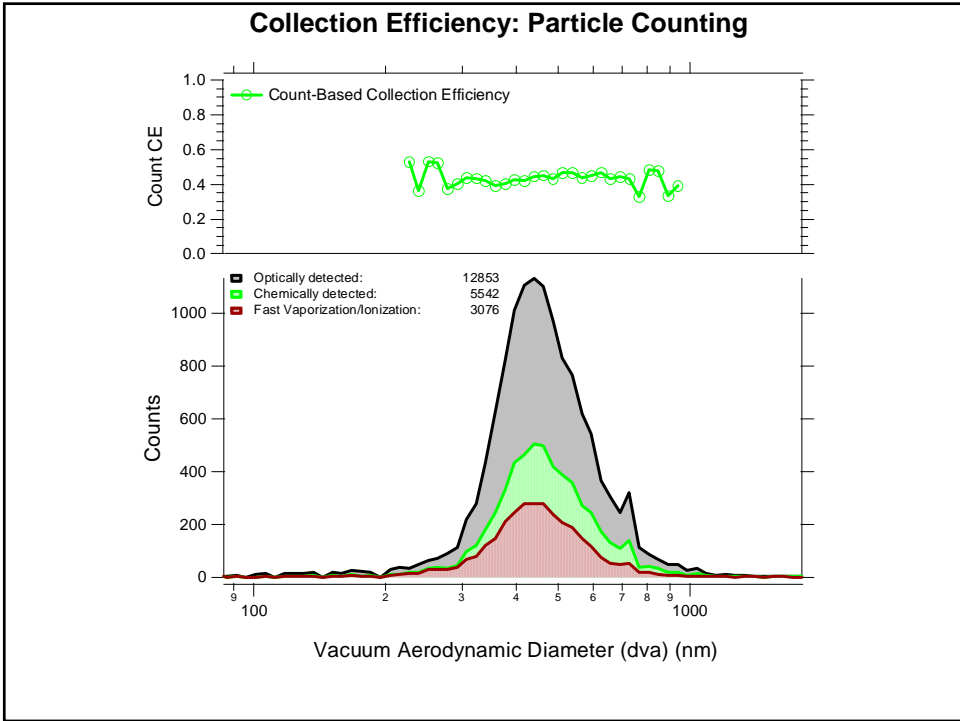
# LS-C-ToF-AMS



## Coincident Particles: Evidence of External Mixing



PARTICLE PROPERTY	Velocity (m/s)	Vacuum Aerodynamic Diameter (nm)	Optical Diameter (nm)	Predicted Physical Diameter (nm)	Chemical Density (g/cc)	Optical Density (g/cc) [dva/dopt]	Measured Total Mass (fg)	Predicted Total Mass (fg) [Chemical Density]	Predicted Total Mass (fg) [Optical Density]
Particle #1							18.2	26.3	29.0
Particle #2							13.0	8.8	15.8



### Chemical composition of summertime aerosol in the Po Valley (Italy), northern Adriatic and Black Sea

J. Crosier,<sup>a\*</sup> J. D. Allan,<sup>a</sup> H. Coe,<sup>a</sup> K. N. Bower,<sup>a</sup> P. Formenti<sup>b</sup> and P. I. Williams<sup>a</sup>

<sup>a</sup> Centre for Atmospheric Science, SEAES, University of Manchester, UK  
<sup>b</sup> LISA, Creteil, France

$$CE = 0.975 - \frac{SO_4^{-2}}{SO_4^{-2} + NO_3^-} 0.582.$$

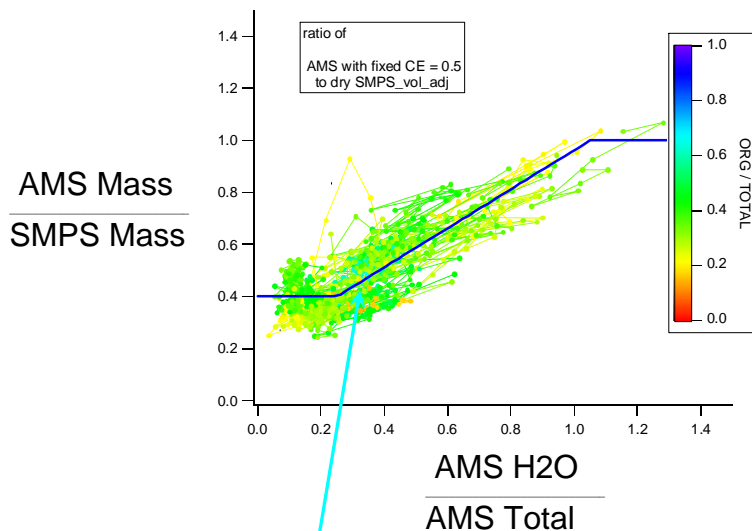
**NH4/SO4 → 0**

**NO3/SO4 → >>1      CE = 0.4 → 1**

**H2O/Total → 1**

**Tim Onasch, Ann Middlebrook: *particle phase***

Junying Sun, Yangmei Zhang Beijing July 2006



Note: cluster of highest org fraction have low H2O and CE

## Lens Transmission issues

Reasonable understanding of current lenses

*Peter Liu, Leah Williams, John Jayne*

*Ann Middlebrook, Brendan Matthew, Ken Docherty*

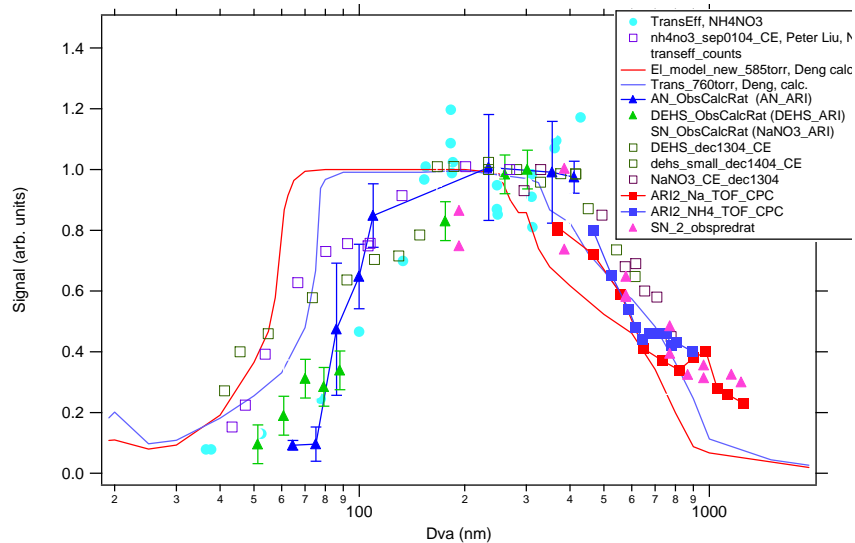
“standard” vs ~~“high throughput”~~

some variability

New “high pressure” lens – modification of Schreiner lens

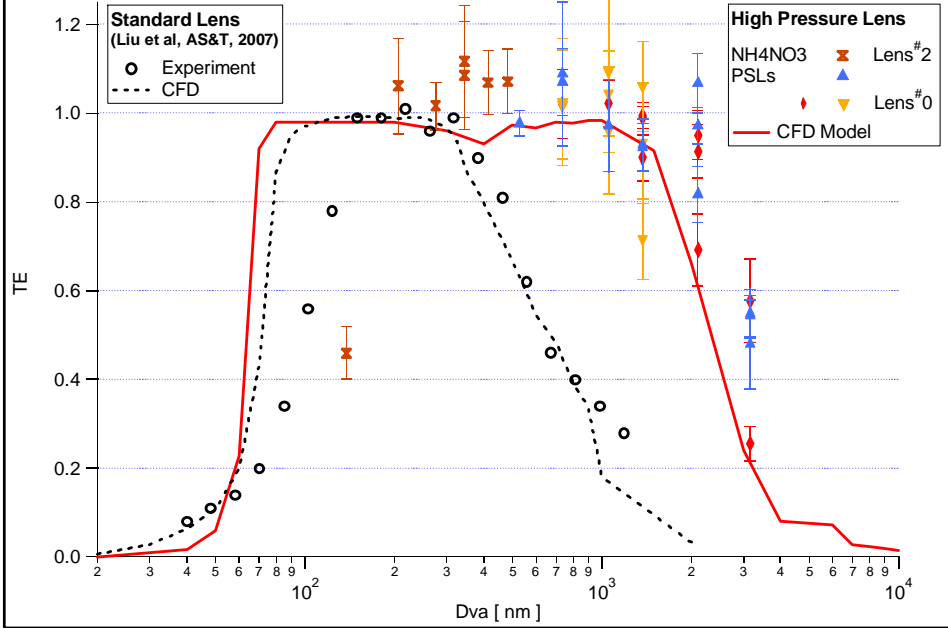
## Summary of Lab Transmission Experiments

“Standard Lens” Williams et al

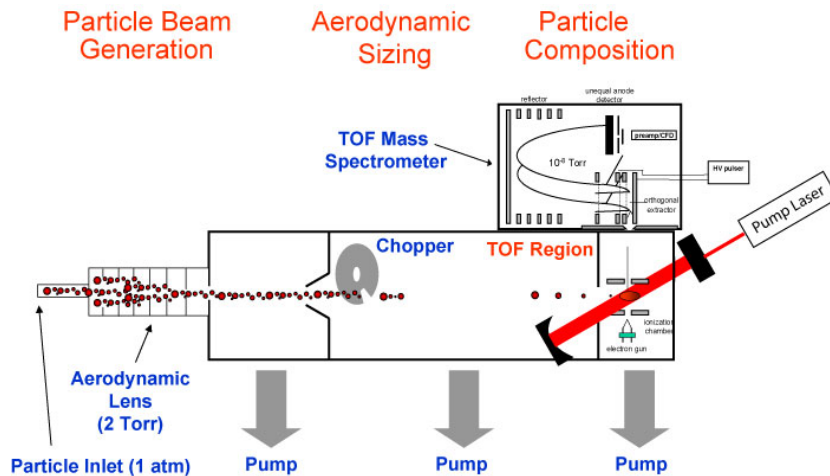




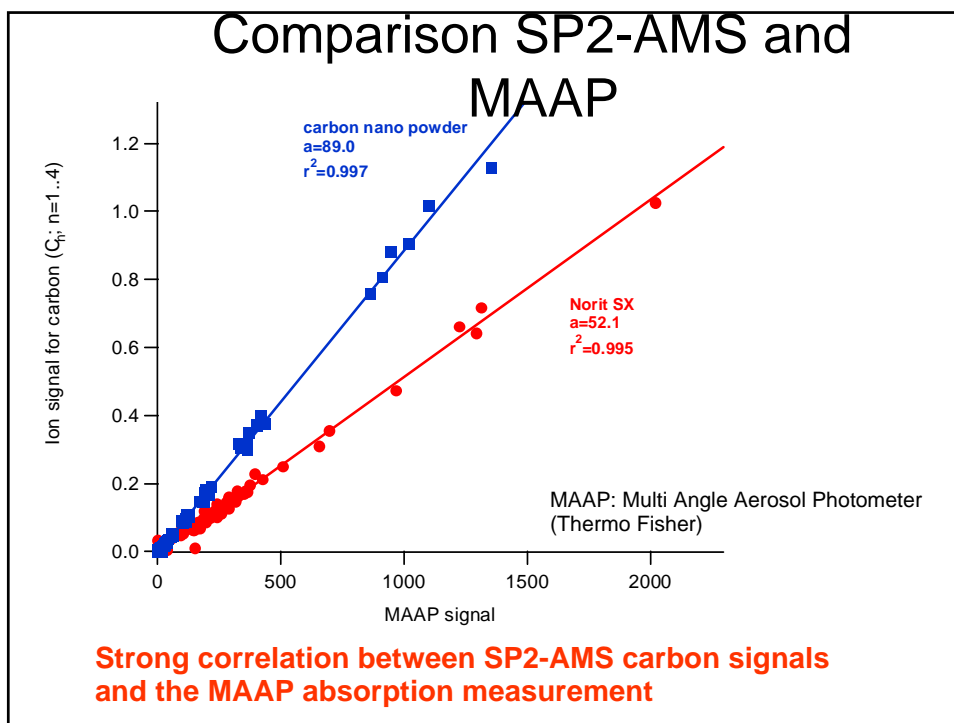
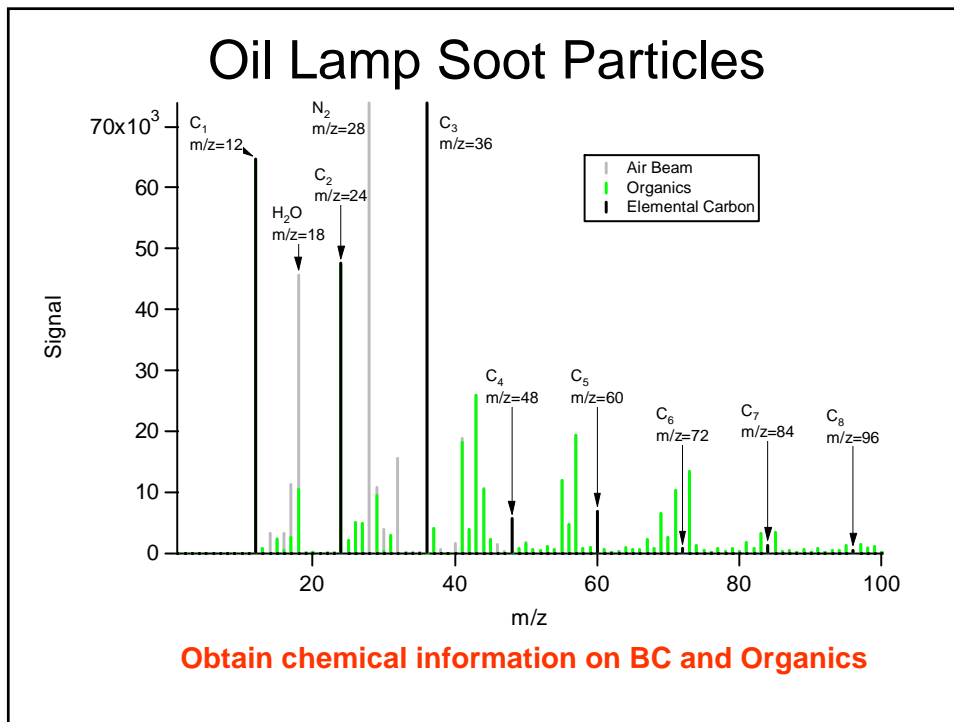
### Comparison of "Standard" and "High Pressure Lenses" August, 2007



### How to build a SP2-AMS



- Install SP2 module



	<b>600C, e-</b>	<b>m/z</b>
HOA:	$C_nH_m \rightarrow C_{n-x}H_{m-y}^+$	27,29,41,43,55,57,69,71,...
OOA2	$C_nH_mO \rightarrow C_2H_3O^+, C_3H_3O^+, R'^+$	43,55, ...
OOA1	$C_nH_mO_2 \rightarrow CO_2^+, HCO_2^+, R'^+$	44,45, ...
BBOA	$R \rightarrow R'^+, C_2H_4O_2^+, C_3H_3O_2^+$	60,73, .... <i>levoglucosan</i>
	SP2 + e-	
Graphitic C	$\rightarrow C_n^+$	

