

# Intro to MS Resolution & Accuracy

CU- Boulder

CHEM-5181

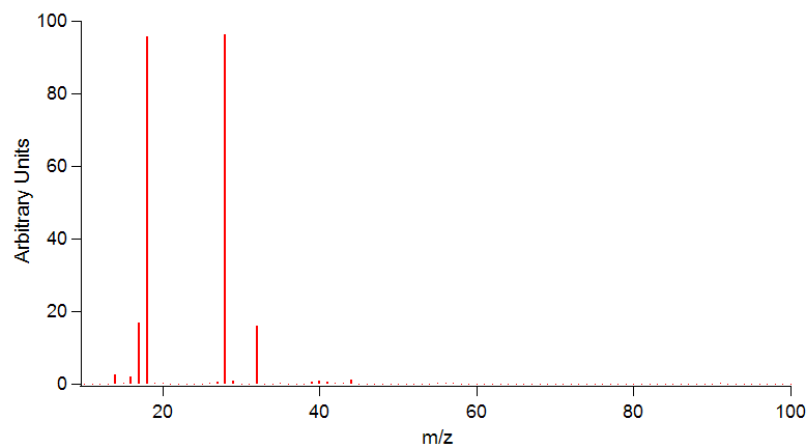
Mass Spectrometry & Chromatography

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*Last updated: Aug 2014*

*Some slides adapted from 2007 lecture by Dr. Joel Kimmel, CU-Boulder*

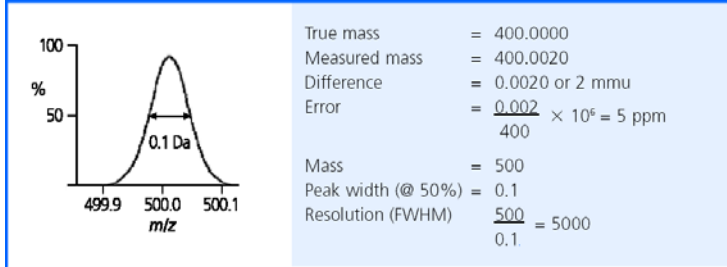
## Mass Spectrum



Signal intensity vs mass-to-charge ratio

# Mass Analyzer Resolving Power

Figure 1: Mass accuracy determination and the FWHM method for determining resolution for a mass spectrometer measured at a given ion.



## Mass peak width ( $\Delta m_{50\%}$ )

Full width of mass spectral peak at half-maximum peak height

## Mass resolution / Resolving Power ( $m / \Delta m_{50\%}$ )

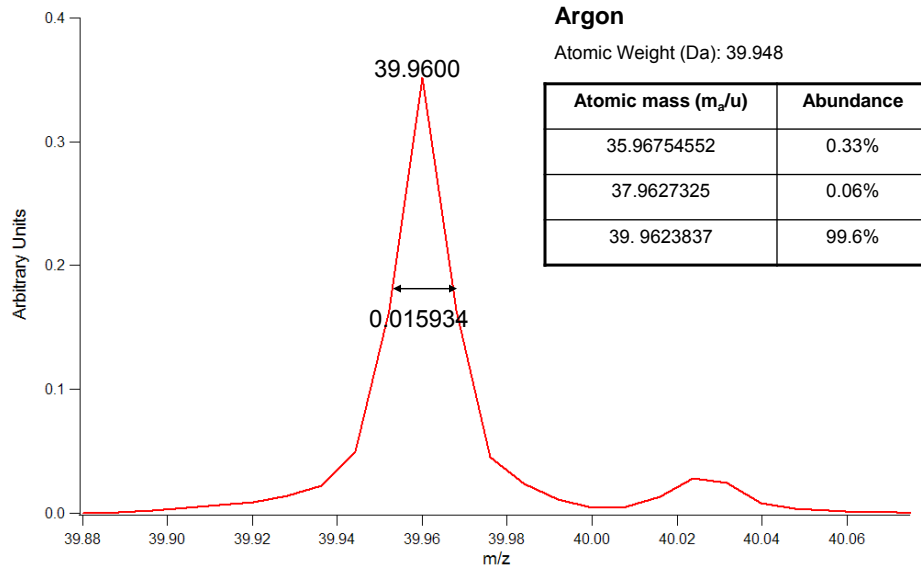
Quantifies ability to isolated single mass spectral peak

## Mass accuracy

Mass accuracy is the difference between measured and actual mass

Figure from: M. P. Balogh, *LC-GC Europe*, 17(3), 152–159 (2004) 3

# Calculate resolution and accuracy



## Mass Analyzer Resolution II

- Question: what is the resolution (FWHM definition) in both cases
  - if  $m_1 = 50$  Da?
  - If  $m_1 = 2500$  Da?

$$(m_2 = m_1 + 1 \text{ amu})$$

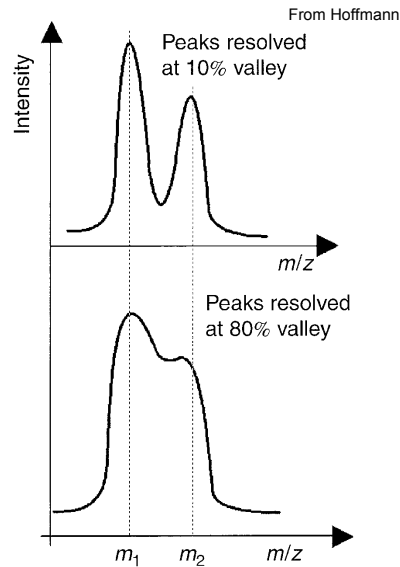


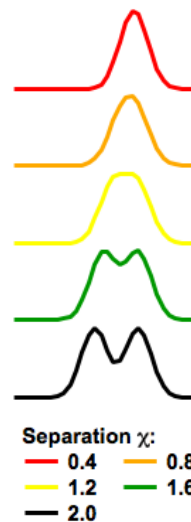
Figure 2.1  
Diagram showing the concepts of peak resolution and valley

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## What Resolution do we really need?

- What resolution is really needed to distinguish these two overlapping ions, *if we know that only these two ions are present?*

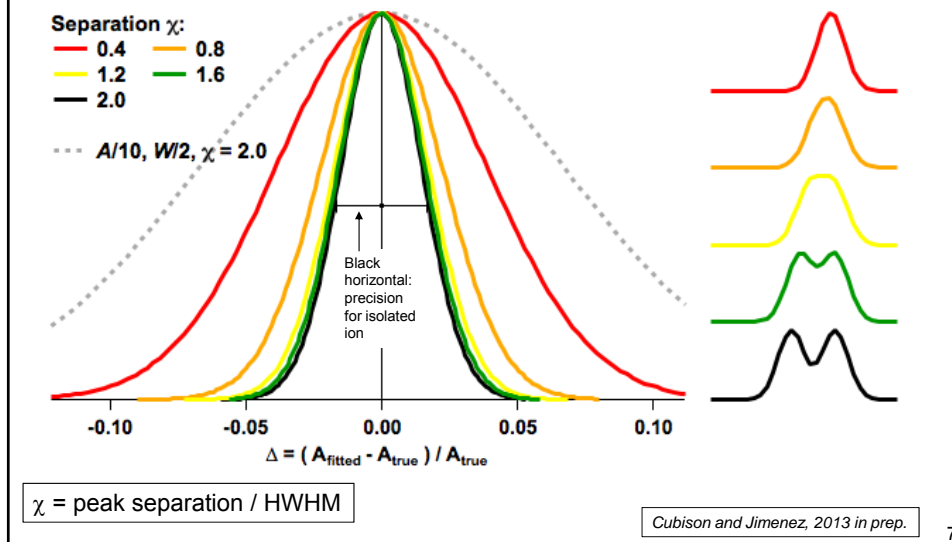
- A.  $\chi = 0.4$
- B.  $\chi = 0.8$
- C.  $\chi = 1.2$
- D.  $\chi = 1.6$
- E.  $\chi = 2.0$



$$\chi = \text{peak separation} / \text{HWHM}$$

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# What Resolution do you really need?



Clicker Q: A mass spectrometer with a resolution of 5000 should be capable of resolving isotopic peaks (e.g.  $^{12}\text{CH}_4$  vs  $^{13}\text{CH}_4$ ) for singly charged species with  $m/z$ :

- (a) Of any value
- (b) Less than  $m/z$  5000
- (c) Greater than  $m/z$  5000
- (d) It depends on the type of mass spectrometer
- (e) I don't know

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